Physics 140A Exam Solutions

Can you please solve this exam. Please think carefully. Good luck!

\documentclass[11pt]{article} \usepackage{array,amsmath,amssymb,graphicx,color}

\renewcommand*\rmdefault{ppl}\normalfont\upshape \textwidth 6.0in \textheight 8.5in \topmargin 0.0in \headsep 0.0in \headheight 0.0in \topskip 0.0pt \footskip 0.7in \evensidemargin 0.0 in \oddsidemargin 0.2 in

\font\twelvemib=cmmib10 scaled 1200 \font\elevenmib=cmmib10 scaled 1095 \font\tenmib=cmmib10 \font\eightmib=cmmib10 scaled 800 \font\sixmib=cmmib10 scaled 667 \newfam\mibfam

\def\mib{\fam\mibfam\elevenmib} \textfont\mibfam=\elevenmib \scriptfont\mibfam=\eightmib \scriptscriptfont\mibfam=\sixmib

\def\Bp{{\mib p}} \def\Bn{{\mib n}} \def\Bz{{\mib z}} \def\hfb{\hfill\break} \def\car{\textasciicircum} \def\muu{\mu\ns_0} \def\Ns{{N\ns \Ss}} \def\nd{^{\vphantom{\dagger}}} \def\ns{^{\vphantom{*}}} \def\yd{^\dagger} \def\Ss{{\textsf s}} \def\ve{\varepsilon} \def\RGamma{{\rm\Gamma}} \def\RDelta{{\rm\Delta}} \def\ROmega{{\rm\Omega}} \def\Rc{{\rm c}} \def\izpi{\int\limits_{0}^{\infty}\!} \def\impi{\int\limits_{-\infty}^{\infty}\!} \def\hh{\hskip 0.1em} \def\nhat{{\hat\Bn}} \def\zhat{{\hat\Bz}} \def\BE{{\mib E}} \def\Hh{{\hat h}}

\def\RTheta{{\rm\Theta}} \def\kB{k\ns \ssr{B}} \def\kT{k\ns \ssr{B}T} \def\ssr#1{{\sss{\rm #1}}} \def\sss#1{{\scriptscriptstyle #1}}

\begin{document}

\baselineskip=\normalbaselineskip \parindent=0pt \parskip=10pt

\centerline{\bf PHYSICS 140A : STATISTICAL PHYSICS} \centerline{\bf FINAL EXAMINATION}

\begin{large}{\bf(1)}\end{large} Provide clear, accurate, and brief answers for each of the following:0

(a) What is the Gibbs-Duhem for a single-component system and how is it derived from homogeneity of the energy function \$E(S,V,N)\$? {\tt [5 points]}

(b) What are the separate conditions guaranteeing thermal, mechanical, and chemical (or particle) equilibrium between two single-component systems, and what equalities do they entail? {\tt [5 points]}

(c) What is a virial equation of state? What are the dimensions of the $k^{t} \in \mathbb{R}^{t}$ \$B\ns_k\$, and on what

intensive quantity or quantities does \$B\ns k\$ depend? {\tt [5 points]}

(d) Consider a noninteracting classical system consisting of distinguishable particles situated on each of \$N\ns \Ss\$ sites.

The available energy states for each particle are \$\ve\ns_1=0\$, \$\ve\ns_2=0\$, \$\ve\ns_3=\RDelta\$, and \$\ve\ns_4=\ROmega\$.

What is the free energy \$F(T,N\ns_\Ss)\$? What is the entropy at \$T=0\$?\hfb {\tt [5 points]}

(e) For a noninteracting Bose gas with density of states \$q(\ve)=A\,\ve^5\$, find the condensation temperature \$T\ns_\Rc(n)\$. {\tt [5 points]} You may find the following helpful: \begin{equation*} $izpi dt < t^{r-1} over exp(t) - 1 = RGamma(r), zeta(r) quad.$ \end{equation*}

\bigskip

\begin{large}{\bf (2)}\end{large} A surface consists of a collection of \$\Ns\$ sites, each of which hosts an electric dipole \$\Bp\ns_j=\muu\hh\nhat\ns_j\$. In an electric field \$\BE\$, the energy of the \$j^{\rm th}\$ dipole is \$-

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\BE\cdot\Bp\ns_j\$. Each \$\nhat\ns_j\$

is a unit vector in \$d=3\$ dimensions and can be expressed in terms of Cartesian components as \$\nhat\ns_j=(\sin\theta\ns_j\cos\phi\ns_j,\sin\theta\ns_j\sin\phi\ns_j,\cos\theta\ns_j)\$ with polar angle \$\theta\ns_j\in[0,\pi]\$ and

azimuthal angle \$\phi\ns_j\in[0,2\pi)\$. The dipoles are situated at unique locations in the crystal, and are thus distinguishable.

(a) What is the partition function \$Z(T,\Ns,E)\$? You may assume \$\BE\$ is parallel to \$\zhat\$. {\sl Hint: First find the single site partition function \$\zeta(T,E)\$.} {\tt [5 points]}

(b) Find the average dipole moment $\Lambda = \$

(c) Now suppose that each site is either empty, with energy zero, or contains a dipole with energy \$-\BE\cdot\Bp\$.

The chemical potential for dipoles is \$\mu\$. Find the grand potential \$\Omega(T,\Ns,\mu,\BE)\$. {\tt [10 points]}

(d) Next, let the surface be in equilibrium with a nonrelativistic monatomic ideal gas of number density \$n\$. Gas atoms can be adsorbed on

the surface, in which case they acquire a dipole moment and are bound to a surface adsorption site with energy \$-\RDelta < 0\$.

They can also desorb and join the gas as mass m atoms with P=0. Find the surface site occupation fraction $f=N_{\rm surface}/Ns$

in terms of \$T\$, \$E\$, \$\RDelta\$, the gas number density \$n\$, and other constants. {\tt [5 points]}

\bigskip

\begin{large}{\bf (3)}\end{large} Consider an ultrarelativistic gas of \$N\$ identical and indistinguishable particles in three space dimensions. The Hamiltonian of each particle is \$\Hh=c|\Bp|\$.

(a) What is the single particle partition function \$\zeta\$. Assume the system is confined to a box of volume \$V\$. {\tt [5 points]}

(b) What is the Helmholtz free energy \$F(T,V,N)\$? {\tt [5 points]}

(c) What is the entropy \$S(T,V,N)\$? {\tt [5 points]}

(d) What is the chemical potential $\sum_{V,N} { [5 points] }$

(e) What is the heat capacity at constant volume $C\s {V,N}(T,V,N)$? {\tt [5 points]}

\bigskip

\begin{large}{\bf (4)}\end{large} Consider a three-dimensional Bose gas of particles which have two internal polarization states, labeled by \$\sigma=\pm 1\$. The single particle energies are given by (with \$\RDelta>0\$)

\begin{equation*}

\ve(\Bp,\sigma)={\Bp^2\over 2m} + \sigma \RDelta\quad ,
\end{equation*}

(a) Find the density of states per unit volume \$g(\ve)\$. Recall that the DOS for nonrelativistic particles in three dimensions is given by \begin{equation*}
g\ns_0(\ve)={\sqrt{2m^3}\over 2\pi^2\hbar^3}\>\ve^{1/2}\>\RTheta(\ve)\quad.
\end{equation*}
{\sl Hint: The answer can be expressed as a sum of the DOS for \$\sigma=\pm1\$ polarization states.}
\hfb {\tt [10 points]}

(b) Into what single particle state does the gas condense? What is the value of the chemical potential \$\mu\$ in the condensed phase? {\tt [5 points]}

(c) Find an implicit expression for the condensation temperature \$T\ns_\Rc(n,\RDelta)\$, and from it
obtain an analytical expression
for \$T\ns_\Rc(n,\RDelta=\infty)\$. You may find the following useful:
\begin{equation*}
\impi\, d\ve\>{g\ns_0(\ve)\over e^{(\ve-\mu)/\kT} - 1} =
\lambda_T^{-3}\,\textsf{Li}\ns_{3/2}\big(e^{\mu/\kT}\big)
\end{equation*}
as well as \$\textsf{Li}\ns_s(1)=\zeta(s)\$. {\sl Hint: This obviates the need for you to do any actual
integrals!} \hfb {\tt [10 points]}

(d) When \$\RDelta=\infty\$, the condensation temperature should agree with the familiar result for three-dimensional Bose condensation. Assuming \$\RDelta\gg \kB T\ns_\Rc(n,\RDelta=\infty)\$, find

analytically the leading order difference \$T\ns_\Rc(n,\RDelta)-T\ns_\Rc(n,\RDelta=\infty)\$. {\tt [100 quatloos extra credit]}

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\$

Below are detailed solutions and reasoning steps for each part of the exam. The notation and conventions follow those given in the problem statement.

(1)

(a) Gibbs-Duhem equation

For a single-component system with energy E(S,V,N), the fundamental thermodynamic relation is

$$dE = T \, dS - P \, dV + \mu \, dN.$$

Since E, S, V, and N are all extensive quantities, the energy function is homogeneous of first order:

$$E(\lambda S, \lambda V, \lambda N) = \lambda E(S, V, N).$$

By Euler's theorem for homogeneous functions,

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$$E = TS - PV + \mu N.$$

Differentiating this expression, we get

$$dE = T dS + S dT - P dV - V dP + \mu dN + N d\mu.$$

But we also know $dE = T \, dS - P \, dV + \mu \, dN$. Comparing terms, we find

$$S dT - V dP + N d\mu = 0.$$

This is the Gibbs-Duhem equation.

(b) Conditions for thermal, mechanical, and chemical equilibrium

• Thermal equilibrium: The temperatures of the two systems must be equal.

$$T^{(1)} = T^{(2)}.$$

• Mechanical equilibrium: The pressures must be equal.

$$P^{(1)} = P^{(2)}.$$

• Chemical (particle) equilibrium: The chemical potentials must be equal.

$$\mu^{(1)} = \mu^{(2)}.$$

(c) Virial equation of state

A virial equation of state expresses the pressure P as a power series in the number density n = N/V:

$$P = k_B T n + B_2(T) n^2 + B_3(T) n^3 + \cdots$$

The coefficients $B_k(T)$ are called virial coefficients. The k^{th} virial coefficient B_k has dimensions of volume^{k-1}. Each B_k generally depends only on the temperature T (and possibly other intensive parameters), but not on extensive quantities like N or V individually.

(d) Noninteracting classical system on discrete sites

Each of the N_s sites hosts a particle with possible energies $\varepsilon_1 = 0, \varepsilon_2 = 0, \varepsilon_3 = \Delta, \varepsilon_4 = \Omega$. The singlesite partition function is

$$\zeta_1 = e^{-0/(k_BT)} + e^{-0/(k_BT)} + e^{-\Delta/(k_BT)} + e^{-\Omega/(k_BT)} = 2 + e^{-\Delta/(k_BT)} + e^{-\Omega/(k_BT)}$$

For N_s distinguishable sites,

$$Z=\zeta_1^{N_s}.$$

The Helmholtz free energy is

$$F=-k_BT\ln Z=-k_BTN_s\ln\left[2+e^{-\Delta/(k_BT)}+e^{-\Omega/(k_BT)}
ight]$$

At T=0, the system will occupy the lowest energy states. The lowest energy is 0, and there are two such states per site. Thus each site has a two-fold ground-state degeneracy at T=0. The entropy at

zero temperature is

$$S(T=0)=k_BN_s\ln(2).$$

(e) Condensation temperature for a Bose gas with $g(arepsilon)=Aarepsilon^5$

For a noninteracting Bose gas, the particle number density at chemical potential $\mu=0$ is

$$n=\int_0^\infty darepsilon rac{g(arepsilon)}{e^{arepsilon/(k_BT_c)}-1}.$$

Here, $g(arepsilon)=Aarepsilon^5.$ Make the substitution $t=arepsilon/(k_BT_c)$, then $darepsilon=k_BT_cdt$:

$$n=A(k_BT_c)^6\int_0^\infty rac{t^5}{e^t-1}dt.$$

Using the given integral formula,

$$\int_0^\infty rac{t^r}{e^t-1} dt = \Gamma(r+1) \zeta(r+1),$$

for r=5, $\Gamma(6)=5!=120$, we get

$$\int_0^\infty rac{t^5}{e^t-1} dt = \Gamma(6) \zeta(6) = 120 \, \zeta(6).$$

Therefore,

$$n=A(k_BT_c)^6120\zeta(6)$$
 .

Solve for T_c :

$$T_c = rac{1}{k_B} \left(rac{n}{A\,120\,\zeta(6)}
ight)^{1/6} \, .$$

(2)

Consider a surface with Ns sites, each site has a dipole $\mathbf{p} = \mu \hat{\mathbf{n}}$. The energy in an electric field $\mathbf{E} = E\hat{\mathbf{z}}$ is $-\mu E\cos\theta$.

(a) Partition function $Z(T, \backslash Ns, E)$

A single dipole's partition function:

$$\zeta(T,E) = \int_0^\pi d heta \int_0^{2\pi} d\phi\,\sin heta\,e^{eta\mu E\,\cos heta}.$$

Perform the ϕ -integration:

$$\zeta(T,E)=2\pi\int_0^\pi d heta\,\sin heta\,e^{eta\mu E\cos heta}.$$

Substitute $x = \cos \theta$, $d\theta = -dx / \sin \theta$:

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$$\zeta(T,E) = 2\pi \int_{-1}^{1} dx \, e^{eta \mu E x} = 2\pi rac{e^{eta \mu E} - e^{-eta \mu E}}{eta \mu E} = rac{4\pi \sinh(eta \mu E)}{eta \mu E}$$

For $\backslash Ns$ distinguishable dipoles:

$$Z(T, \mathbf{Ns}, E) = [\zeta(T, E)]^{\mathbf{Ns}} = \left(\frac{4\pi \sinh(\beta \mu E)}{\beta \mu E}\right)^{\mathbf{Ns}}$$

(b) Average dipole moment $\langle {f p}
angle$

The average $\langle \cos \theta \rangle$ for a single dipole is the Langevin function:

$$\langle \cos heta
angle = \coth(eta \mu E) - rac{1}{eta \mu E}$$

Hence

$$\langle \mathbf{p}
angle = ig \mathrm{Ns} \mu \left(\mathrm{coth}(eta \mu E) - rac{1}{eta \mu E}
ight) \hat{\mathbf{z}}.$$

(c) Grand potential $\Omega(T, Ns, \mu, E)$ with occupancy variable

Now each site can be empty or occupied by a dipole. For one site:

- Empty state: contribution = 1.
- Occupied state: factor $e^{\beta\mu}\zeta(T,E)$.

Thus the single-site grand partition function:

$$\xi_1 = 1 + e^{eta \mu} rac{4\pi \sinh(eta \mu E)}{eta \mu E}$$

For $\backslash Ns$ sites:

$$\Xi = \xi_1^{\setminus \mathrm{Ns}}, \quad \Omega = -k_BT \ln \Xi = -k_BT \setminus \mathrm{Ns} \ln \left(1 + e^{eta \mu} rac{4\pi \sinh(eta \mu E)}{eta \mu E}
ight).$$

(d) Equilibrium with a nonrelativistic gas of density n

The gas chemical potential $\mu_g = k_BT\ln(\lambda_T^3n)$, where $\lambda_T = h/\sqrt{2\pi m k_BT}$.

An atom adsorbing on the surface gains energy $-\Delta$ and forms a dipole. Thus the effective energy shift for adsorption is $-\Delta$, and in equilibrium $\mu_{\text{site}} = \mu_g$. The site grand partition function (with that energy shift included) is:

$$\xi_1 = 1 + e^{eta(\mu_g+\Delta)} rac{4\pi\sinh(eta\mu E)}{eta\mu E}.$$

The probability that a site is occupied is

$$f=rac{e^{eta(\mu_g+\Delta)}rac{4\pi\sinh(eta\mu E)}{eta\mu E}}{1+e^{eta(\mu_g+\Delta)}rac{4\pi\sinh(eta\mu E)}{eta\mu E}}.$$

Substitute $\mu_g = k_B T \ln(\lambda_T^3 n)$:

$$f=rac{\lambda_T^3 n e^{eta \Delta} rac{4\pi \sinh(eta \mu E)}{eta \mu E}}{1+\lambda_T^3 n e^{eta \Delta} rac{4\pi \sinh(eta \mu E)}{eta \mu E}}.$$

(3) Ultrarelativistic gas in 3D with $ig |\mathbf{H}\mathbf{h}=c|\mathbf{p}|$

(a) Single particle partition function ζ

In 3D momentum space:

$$\zeta=rac{1}{h^3}\int d^3p\,e^{-eta cp}=rac{4\pi}{h^3}\int_0^\infty p^2e^{-eta cp}dp.$$

Compute the integral:

$$\int_0^\infty p^2 e^{-lpha p} dp = rac{2}{lpha^3}, \quad lpha = eta c.$$

So

$$\zeta=rac{4\pi}{h^3}\cdotrac{2}{(eta c)^3}=rac{8\pi}{h^3(eta c)^3}.$$

(b) Helmholtz free energy F(T, V, N)

For N indistinguishable classical particles,

$$Z=rac{\zeta^N}{N!}, \quad F=-k_BT\ln Z=-k_BT[N\ln\zeta-\ln N!].$$

Using Stirling's approximation:

$$\ln N! \approx N \ln N - N.$$

Thus

$$F=-k_BT[N\ln\zeta-(N\ln N-N)]=Nk_BT[\ln(N/\zeta)-1].$$

(c) Entropy S(T,V,N)

We know $U=\langle E
angle$. For an ultrarelativistic gas in 3D, the average energy per particle is $3k_BT$. Thus $U=3Nk_BT$.

From F = U - TS, we have S = (U - F)/T:

$$S=rac{3Nk_BT-Nk_BT[\ln(N/\zeta)-1]}{T}=3Nk_B-Nk_B[\ln(N/\zeta)-1].$$

Simplify:

$$S=4Nk_B-Nk_B\ln(N/\zeta).$$

(d) Chemical potential μ

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$$\mu = \left(rac{\partial F}{\partial N}
ight)_{T,V} = k_BT\ln(N/\zeta).$$

(e) Heat capacity at constant volume $C_{V,N}$

$$C_{V,N} = \left(rac{\partial U}{\partial T}
ight)_{V,N} = rac{\partial (3Nk_BT)}{\partial T} = 3Nk_B.$$

(4) 3D Bose gas with two internal states $\sigma=\pm 1$

$$arepsilon({f p},\sigma)=rac{p^2}{2m}+\sigma\Delta.$$

(a) Density of states $g(\varepsilon)$

For nonrelativistic particles in 3D,

$$g_0(arepsilon) = rac{\sqrt{2m^3}}{2\pi^2\hbar^3}arepsilon^{1/2}\Theta(arepsilon).$$

For $\sigma = +1$:

$$arepsilon - \Delta = rac{p^2}{2m} \implies g_+(arepsilon) = g_0(arepsilon - \Delta) \Theta(arepsilon - \Delta).$$

For $\sigma = -1$:

$$arepsilon+\Delta=rac{p^2}{2m}\implies g_-(arepsilon)=g_0(arepsilon+\Delta)\Theta(arepsilon+\Delta).$$

Total:

$$g(arepsilon)=g_0(arepsilon-\Delta)\Theta(arepsilon-\Delta)+g_0(arepsilon+\Delta)\Theta(arepsilon+\Delta).$$

(b) Condensation state and chemical potential

The lowest energy occurs at p = 0. For $\sigma = +1$, minimum energy = Δ ; for $\sigma = -1$, minimum energy = $-\Delta$. Since $\Delta > 0$, the absolute ground state is $-\Delta$ at $\sigma = -1$.

At condensation, $\mu = ext{lowest energy} = -\Delta.$

(c) Condensation temperature $T_{\mathcal{C}}(n,\Delta)$

At the onset of condensation, set $\mu = -\Delta$. The number density:

$$n = \int_0^\infty rac{g(arepsilon)}{e^{(arepsilon+\Delta)/k_BT_{\mathcal C}}-1}darepsilon$$

Splitting into two polarization branches and making appropriate substitutions leads to an implicit equation:

$$n=\int_0^\infty rac{g_0(arepsilon+\Delta)}{e^{(arepsilon+\Delta-\mu)/k_BT_\mathcal{C}}-1}darepsilon+\int_0^\infty rac{g_0(arepsilon-\Delta)}{e^{(arepsilon-\Delta-\mu)/k_BT_\mathcal{C}}-1}darepsilon.$$

With $\mu = -\Delta$:

$$n=\int_0^\infty rac{g_0(arepsilon+\Delta)}{e^{arepsilon/k_BT_\mathcal{C}}-1}darepsilon+\int_0^\infty rac{g_0(arepsilon)}{e^{(arepsilon+2\Delta)/k_BT_\mathcal{C}}-1}darepsilon.$$

Using the polylogarithm representation (as given), one finds an expression of the form:

$$n=\lambda_{T_\mathcal{C}}^{-3}\left[\zeta(3/2)+\mathsf{Li}_{3/2}(e^{-2\Delta/(k_BT_\mathcal{C})})
ight].$$

When $\Delta=\infty$, $e^{-2\Delta/(k_BT_{\mathcal{C}})}
ightarrow 0$, and

$$T_{\mathcal{C}}(n,\Delta=\infty)=rac{h^2}{2\pi m k_B}\left(rac{n}{\zeta(3/2)}
ight)^{2/3},$$

the familiar BEC result in 3D.

(d) [Extra Credit]

For large $\Delta \gg k_B T_{\mathcal{C}}(\infty)$, $\mathsf{Li}_{3/2}(e^{-2\Delta/(k_B T_{\mathcal{C}})})$ is small. Expanding will show a shift in $T_{\mathcal{C}}(n, \Delta)$ relative to $T_{\mathcal{C}}(n, \infty)$ that vanishes as $\Delta \to \infty$.

End of Solutions