

energies for electrons in 2D square box of length L :
 (n_1, n_2) $n_1^2 + n_2^2$

$$E_{n_1, n_2} = \frac{\hbar^2 \pi^2}{2m_e L^2} (n_1^2 + n_2^2)$$

$(1,3)$ \equiv 10
 $(3,1)$

$(2,2)$ \uparrow 8

$(1,2)$ $\uparrow \downarrow$ 5
 $(2,1)$ $\uparrow \downarrow$

$(1,1)$ $\uparrow \downarrow$ 2

Lowest energy states are:

$(1,1)$ $(1,2)$ $(2,2)$ $(3,1)$ $(3,2)$
 $(2,1)$ $(1,3)$ $(2,3)$

According to the Pauli exclusion principle, we can have 2 electrons of opposite spin in each energy state. See diagram, for 7 electrons

The lowest excitation energy is for the electron in $(2,2)$ to go to $(3,1)$ or $(1,3)$, with energy change

$$\Delta E = E_{3,1} - E_{2,2} = \frac{\hbar^2 \pi^2}{2m_e L^2} (10 - 8) = \frac{\hbar^2 \pi^2}{2m_e L^2} \cdot 2 \Rightarrow$$

$$\Delta E = \frac{7.62 \pi^2}{L^2} = \frac{hc}{\lambda} \Rightarrow L^2 = \frac{7.62 \pi^2 \lambda}{hc} = 25 \text{ \AA}^2 \Rightarrow$$

$$\Rightarrow \boxed{L = 5 \text{ \AA}} \quad (a)$$

(b) The next lowest excitation energy is for an electron in $(1,2)$ or $(2,1)$ to go to $(2,2)$, with energy change

$$\Delta E = E_{2,2} - E_{1,2} = \frac{\hbar^2 \pi^2}{2m_e L^2} \cdot 3 = \frac{hc}{\lambda} \Rightarrow \lambda = \frac{2}{3} \times 4122 \text{ \AA}$$

$$\Rightarrow \boxed{\lambda = 2748 \text{ \AA}}$$

Problem 2

$$\Psi(r, \theta, \phi) = C r^2 e^{-r/a_0} \sin\theta \cos\theta e^{-i\phi}$$

From $e^{-i\phi} \Rightarrow m_l = -1$

From r^2 factn $\Rightarrow n=3$ and $l=2$

From $e^{-2r/na_0} = e^{-r/a_0} \Rightarrow z=3$

(b) $P(r) = r^2 R(r)^2 \propto r^6 e^{-2r/a_0}$

$$P'(r) = 0 = 6r^5 - \frac{2r^6}{a_0} \Rightarrow \boxed{r_m = 3a_0}$$

According to Bohr atom, $r_n = \frac{a_0}{Z} n^2 = \frac{a_0}{3} \cdot 3^2 = 3a_0 = r_m$

\Rightarrow they agree.

(c) Electrons have spin $S = 1/2$, the total angular momentum quantum number j satisfies

$$|l-s| \leq j \leq l+s$$

$$\Rightarrow j = 3/2 \text{ or } j = 5/2$$

Possible values of m_j : $-j$ to j in integer steps \Rightarrow

$$-\frac{5}{2}, -\frac{3}{2}, -\frac{1}{2}, \frac{1}{2}, \frac{3}{2}, \frac{5}{2}$$

Problem 3

The average energy of a 1D harmonic oscillator is

$$E = \frac{\hbar\omega}{2} + \frac{\hbar\omega}{e^{\hbar\omega/k_B T} - 1}$$

Ground state energy is $E_0 = \frac{\hbar\omega}{2} = 0.01 \text{ eV} \Rightarrow \boxed{\hbar\omega = 0.02 \text{ eV}}$

For $E = 0.02 \text{ eV} = \hbar\omega \Rightarrow$

$$\frac{\hbar\omega}{2} + \frac{\hbar\omega}{e^{\hbar\omega/k_B T} - 1} = \hbar\omega \Rightarrow \frac{1}{e^{\hbar\omega/k_B T} - 1} = \frac{1}{2} \Rightarrow$$

$$\Rightarrow e^{\hbar\omega/k_B T} = 3 \Rightarrow T = \frac{\hbar\omega}{k_B \ln 3} = \boxed{211.2 \text{ K}} \quad (a)$$

Note: the "Einstein temperature" is $T_E = \frac{\hbar\omega}{k_B} = 232 \text{ K}$

(b) According to equipartition,

$$E = k_B T = \frac{211.2}{11,600} \text{ eV} = \boxed{0.0182 \text{ eV}}$$

it is lower because it does not take into account zero-point energy

(c) The heat capacity for 1 oscillator is

$$C = \frac{dE}{dT} = \left(\frac{\hbar\omega}{k_B T}\right)^2 \frac{e^{\hbar\omega/k_B T}}{(e^{\hbar\omega/k_B T} - 1)^2} k_B = \frac{(\ln 3)^2 e^{\ln 3}}{(e^{\ln 3} - 1)^2} k_B =$$

$$= \frac{3}{4} (\ln 3)^2 k_B = \boxed{0.905 k_B}$$

equipartition predicts $C = k_B$, so it differs by $\boxed{\sim 10\%}$
(9.5% more precisely)

Exact answer is lower than predicted by equipartition at all T.