6-44. (a) For
$$x > 0$$
, $\hbar^2 k_2^2 / 2m + V_0 = E = \hbar^2 k_1^2 / 2m = 2V_0$
So, $k_2 = 2mV_0^{-1/2}/\hbar$. Because $k_1 = 4mV_0^{-1/2}/\hbar$, then $k_2 = k_1 / \sqrt{2}$

(b)
$$R = k_1 - k_2^2 / k_1 + k_2^2$$
 (Equation 6-68)
= $1 - 1/\sqrt{2}^2 / 1 + 1/\sqrt{2}^2 = 0.0294$, or 2.94% of the incident particles are reflected.

- (c) T = 1 R = 1 0.0294 = 0.971
- (d) 97.1% of the particles, or $0.971 \times 10^6 = 9.71 \times 10^5$, continue past the step in the +x direction. Classically, 100% would continue on.

6-45 (a) Equation 6-76:
$$T \approx 16 \frac{E}{V_0} \left(1 - \frac{E}{V_0} \right) e^{-2\alpha a}$$
 where $\alpha = 2\sqrt{2m_p(V_0 - E)} / \hbar$

and a =barrier width.

$$-2\alpha a = -2\left[\sqrt{2(938 \text{MeV/c}^2)(50 - 44) \text{MeV}} / 6.58 \times 10^{-22} \text{ MeV} \cdot \text{s}\right] \times 10^{-15} = -1.075$$

$$T \approx 16 \frac{44 \text{ MeV}}{50 \text{ MeV}} \left(1 - \frac{44 \text{ MeV}}{50 \text{ MeV}}\right) e^{-1.075}$$

$$T \approx 0.577$$

(b) decay rate $\approx N \times T$ where

$$N = \frac{v_{proton}}{2R} = \left[\frac{2 \times 44 \,\text{MeV} \times 1.60 \times 10^{-13} \,\text{J/MeV}}{1.67 \times 10^{-27} \,\text{kg}} \right]^{1/2} \times \frac{1}{2 \times 10^{-15} \,\text{m}} = 4.59 \times 10^{22} \,\text{s}^{-1}$$

$$\text{decay rate} \approx 0.577 \times 4.59 \times 10^{22} \,\text{s}^{-1} = 2.65 \times 10^{22} \,\text{s}^{-1}$$

(c) In the expression for T, $e^{-1.075} \Rightarrow e^{-2.150}$, and so $T \approx 0.577 \Rightarrow T \approx 0.197$. The decay rate then becomes $9.05 \times 10^{21} \,\text{s}^{-1}$, a factor of $0.34 \times$ the original value.

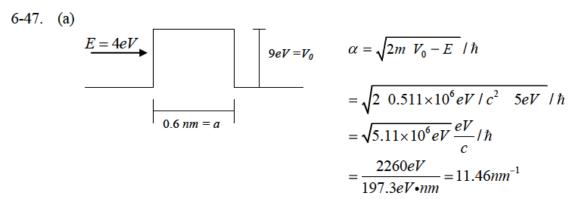
6-46. (a) For
$$x > 0$$
, $\hbar^2 k_2^2 / 2m - V_0 = E = \hbar^2 k_1^2 / 2m = 2V_0$
So, $k_2 = \frac{6mV_0^{-1/2}}{\hbar}$. Because $k_1 = \frac{4mV_0^{-1/2}}{\hbar}$, then $k_2 = \sqrt{3/2}k_1$

(b)
$$R = k_1 - k_2^2 / k_1 + k_2^2$$

 $R = k_1 - k_2^2 / k_1 + k_2^2 = 1 - \sqrt{3/2}^2 / 1 + \sqrt{3/2}^2 = 0.0102$

Or 1.02% are reflected at x = 0.

- (c) T = 1 R = 1 0.0102 = 0.99
- (d) 99% of the particles, or $0.99 \times 10^6 = 9.9 \times 10^5$, continue in the +x direction. Classically, 100% would continue on.



and $\alpha a = 0.6nm \times 11.46nm^{-1} = 6.87$

Since αa is not $\ll 1$, use Equation 6-75:

The transmitted fraction

$$T = \left[1 + \frac{\sinh^2 \alpha a}{4 E/V_0 1 - E/V_0} \right]^{-1} = \left[1 + \left(\frac{81}{80} \right) \sinh^2 6.87 \right]^{-1}$$

Recall that $\sinh x = e^x - e^{-x} / 2$,

$$T = \left[1 + \frac{81}{80} \left(\frac{e^{6.87} - e^{-6.87}}{2} \right)^2 \right]^{-1} = 4.3 \times 10^{-6} \text{ is the transmitted fraction.}$$

(b) Noting that the size of T is controlled by αa through the $\sinh^2 \alpha a$ and increasing T implies increasing E. Trying a few values, selecting E = 4.5eV yields $T = 8.7 \times 10^{-6}$ or approximately twice the value in part (a).

6-50. Using Equation 6-76,

$$T \approx 16 \frac{E}{V_0} \left(1 - \frac{E}{V_0} \right) e^{-2\alpha a} \text{ where } E = 2.0 eV, \ V_0 = 6.5 eV, \text{ and } a = 0.5 nm.$$

$$T \approx 16 \left(\frac{2.0}{6.5} \right) \left(1 - \frac{2.0}{6.5} \right) e^{-2.10.87 - 0.5} \approx 6.5 \times 10^{-5} \text{ (Equation 6-75 yields } T = 6.6 \times 10^{-5}.)$$

6-51.
$$R = \frac{k_1 - k_2^2}{k_1 + k_2^2}$$
 and $T = 1 - R$ (Equations 6-68 and 6-70)

(a) For protons:

(b) For electrons:

$$k_1 = 1.388 \left(\frac{0.511}{938}\right)^{1/2} = 0.0324$$
 $k_2 = 0.694 \left(\frac{0.511}{938}\right)^{1/2} = 0.0162$ $R = \left(\frac{0.0324 - 0.0162}{0.0324 + 0.0162}\right)^2 = 0.111$ And $T = 1 - R = 0.889$

No, the mass of the particle is not a factor. (We might have noticed that \sqrt{m} could be canceled from each term.

6-56. (a) The requirement is that $\psi^2 = x = \psi^2 - x = \psi - x = \psi - x$. This can only be true if: $\psi - x = \psi = x$ or $\psi - x = -\psi = x$.

(b) Writing the Schrödinger equation in the form $\frac{d^2\psi}{dx^2} = -\frac{2mE}{\hbar^2}\psi$, the general solutions of this 2nd order differential equation are: $\psi x = A\sin kx$ and $\psi x = A\cos kx$ where $k = \sqrt{2mE}/\hbar$. Because the boundaries of the box are at $x = \pm L/2$, both

solutions are allowed (unlike the treatment in the text where one boundary was at

$$x=0$$
). Still, the solutions are all zero at $x=\pm L/2$ provided that an integral number of half wavelengths fit between $x=-L/2$ and $x=+L/2$. This will occur for: $\psi_n \ x = 2/L^{1/2} \cos n\pi x/L$ when $n=1,3,5,\cdots$. And for $\psi_n \ x = 2/L^{1/2} \sin n\pi x/L$ when $n=2,4,6,\cdots$.

The solutions are alternately even and odd.

(c) The allowed energies are: $E = h^2 k^2 / 2m = h^2 n\pi / L^2 / 2m = n^2 h^2 / 8mL^2$.

6-58. $\langle x^2 \rangle = \int_{L}^{L} x^2 \sin^2 \frac{n\pi x}{L} dx$ Letting $u = n\pi x/L$, $du = n\pi/L dx$

 $= \frac{2}{L} \left(\frac{L}{n\pi} \right)^3 \left| \frac{u^3}{6} - \left(\frac{u^2}{4} - \frac{1}{8} \right) \sin 2u - \frac{u \cos 2u}{4} \right|^{n\pi}$

 $= \frac{2}{L} \left(\frac{L}{n\pi} \right)^{3} \left| \frac{n\pi^{3}}{6} - 0 - \frac{n\pi}{4} - 0 \right| = \frac{L^{2}}{3} - \frac{L^{2}}{2n^{2}\pi^{2}}$

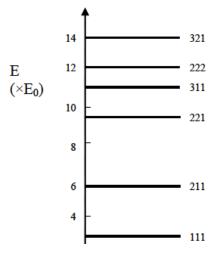
 $\langle x^2 \rangle = \frac{2}{L} \left(\frac{L}{n\pi} \right)^2 \left(\frac{L}{n\pi} \right) \int_{0}^{n\pi} u^2 \sin^2 u du$

7-1.
$$E_{n_1 n_2 n_3} = \frac{\hbar^2 \pi^2}{2mL^2} (n_1^2 + n_2^2 + n_3^2)$$
 (Equation 7-4)

$$E_{311} = \frac{\hbar^2 \pi^2}{2mL^2} (3^2 + 1^2 + 1^2) = 11E_0$$
 where $E_0 = \frac{\hbar^2 \pi^2}{2mL^2}$

$$E_{222} = E_0(2^2 + 2^2 + 2^2) = 12E_0$$
 and $E_{321} = E_0(3^2 + 2^2 + 1^2) = 14E_0$

The 1st, 2nd, 3rd, and 5th excited states are degenerate.



7-2.
$$E_{n_1 n_2 n_3} = \frac{\hbar^2 \pi^2}{2m} \left(\frac{n_1^2}{L_1^2} + \frac{n_2^2}{L_2^2} + \frac{n_3^2}{L_3^2} \right) = \frac{\hbar^2 \pi^2}{2m L_1^2} \left(n_1^2 + \frac{n_2^2}{4} + \frac{n_3^2}{9} \right)$$
 (Equation 7-5)

 $n_1 = n_2 = n_3 = 1$ is the lowest energy level.

$$E_{111} = E_0 (1 + 1/4 + 1/9) = 1.361 E_0 \text{ where } E_0 = \frac{\hbar^2 \pi^2}{2mL_1^2}$$

The next nine levels are, increasing order,

n_1	n_2	n_3	$E\left(\times E_{0}\right)$
1	1	2	1.694
1	2	1	2.111
1	1	3	2.250
1	2	2	2.444
1	2	3	3.000
1	1	4	3.028
1	3	1	3.360
1	3	2	3.472
1	2	4	3.778

7-3. (a)
$$\psi_{n_1 n_2 n_3}(x, y, z) = A \cos \frac{n_1 \pi x}{L} \sin \frac{n_2 \pi y}{L} \sin \frac{n_3 \pi z}{L}$$

(b) They are identical. The location of the coordinate origin does not affect the energy level structure.

7-4.
$$\psi_{111}(x,y,z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{2L_1} \sin \frac{\pi z}{3L_1} \qquad \psi_{112}(x,y,z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{2L_1} \sin \frac{2\pi z}{3L_1}$$

$$\psi_{121}(x,y,z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{L_1} \sin \frac{\pi z}{3L_1} \qquad \psi_{122}(x,y,z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{L_1} \sin \frac{2\pi z}{3L_1}$$

$$\psi_{113}(x,y,z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{2L_1} \sin \frac{\pi z}{L_1}$$

$$\psi_{113}(x,y,z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{2L_1} \sin \frac{\pi z}{L_1}$$

7-7.
$$E_{0} = \frac{\hbar^{2}\pi^{2}}{2mL^{2}} = \frac{\left(1.055 \times 10^{-34} J \cdot s\right)^{2} \pi^{2}}{2\left(9.11 \times 10^{-31} kg\right) \left(0.10 \times 10^{-9} m\right)^{2} \left(1.609 \times 10^{-19} J/eV\right)} = 37.68eV$$

$$E_{311} - E_{111} = \Delta E = 11E_{0} - 3E_{0} = 8E_{0} = 301eV$$

$$E_{222} - E_{111} = \Delta E = 12E_{0} - 3E_{0} = 9E_{0} = 339eV$$

$$E_{321} - E_{111} = \Delta E = 14E_{0} - 3E_{0} = 11E_{0} = 415eV$$

7-8. (a) Adapting Equation 7-3 to two dimensions (i.e., setting $k_3 = 0$), we have

$$\psi_{n_1 n_2} = A \sin \frac{n_1 \pi x}{L} \sin \frac{n_2 \pi y}{L}$$

(b) From Equation 7-5, $E_{n_1 n_2} = \frac{\hbar^2 \pi^2}{2mL^2} (n_1^2 + n_2^2)$

 $n_2 = 1$.

(c) The lowest energy degenerate states have quantum numbers $n_1 = 1$, $n_2 = 2$, and $n_1 = 2$,

7-5.
$$E_{n_1 n_2 n_3} = \frac{\hbar^2 \pi^2}{2m} \left(\frac{n_1^2}{L_1^2} + \frac{n_2^2}{\left(2L_1\right)^2} + \frac{n_3^2}{\left(4L_1\right)^2} \right) = \frac{\hbar^2 \pi^2}{2mL_1^2} \left(n_1^2 + \frac{n_2^2}{4} + \frac{n_3^2}{16} \right) \quad \text{(from Equation 7-5)}$$

$$E_{n_1 n_2 n_3} = \left(n_1^2 + \frac{n_2^2}{4} + \frac{n_3^2}{16} \right) \quad \text{where } E_0 = \frac{\hbar^2 \pi^2}{2mL_1^2}$$

158

Chapter 7 - Atomic Physics

(Problem 7-5 continued)

(a)

n_1	<i>n</i> ₂	<i>n</i> ₃	$E\left(\times E_{0}\right)$
1	1	1	1.313
1	1	2	1.500
1	1	3	1.813
1	2	1	2.063
1	1	4	2.250
1	2	2	2.250
1	2	3	2.563
1	1	5	2.813
1	2	4	3.000
1	3	1	3.313

(b) 1,1,4 and 1,2,2