

Problem 1

(a) The K_{α} photons result from transitions from $n=2$ to $n=1$, so

$$\frac{hc}{\lambda} = E_0(z-1)^2 \left(1 - \frac{1}{4}\right), \quad E_0 = 13.6 \text{ eV} \quad \Rightarrow$$

$$(z-1)^2 = \frac{hc}{\lambda E_0} \cdot \frac{4}{3} \Rightarrow z = 1 + \left(\frac{12,400}{0.4328 \times 13.6} \times \frac{4}{3} \right)^{1/2} \Rightarrow \boxed{z = 54}$$

(b) For Compton scattering,

$$\lambda' = \lambda + \frac{hc}{m_e c} (1 - \cos \theta)$$

largest wavelength for $\theta = \pi \Rightarrow \lambda' = \lambda + \frac{2h}{m_e c} \Rightarrow \boxed{\lambda' = 0.2485 \text{ \AA}}$

smallest wavelength for $\theta = 0 \Rightarrow \boxed{\lambda' = 0.2 \text{ \AA}}$

(c) $E_{\text{kin}}^{\text{electron}} = -\frac{hc}{\lambda'(\theta=\pi)} + \frac{hc}{\lambda} = 12,107 \text{ eV}$

Taking into account the work function, the kinetic energy is $\boxed{12,103 \text{ eV}}$

(d) The highest possible energy is in a process like in the photoelectric effect, so

$$E_{\text{kin}} = \frac{hc}{\lambda} - \phi = 62,000 \text{ eV} - 4 \text{ eV} = \boxed{61,996 \text{ eV}}$$

(e) If the incoming photon is absorbed by an electron in the $n=1$ shell and that electron is ejected, a K_{α} emission results as discussed in (a). The ejected electron has kinetic energy

$$E_{\text{kin}} = \frac{hc}{\lambda} - E_0 \frac{z^2}{n^2} = 62,000 \text{ eV} - 39,657.6 \text{ eV} = 22,342 \text{ eV}$$

or more accurately, taking into account ϕ , $\boxed{E_{\text{kin}} = 22,338.4 \text{ eV}}$

Problem 2

$$(a) \quad -\frac{\hbar^2}{2m} \frac{d^2\psi}{dx^2} + V(x)\psi = E\psi \quad ; \quad \psi = Ae^{-\lambda x^2}$$

$$\frac{d\psi}{dx} = -2\lambda x e^{-\lambda x^2}; \quad \frac{d^2\psi}{dx^2} = -2\lambda e^{-\lambda x^2} + 4\lambda^2 x^2 e^{-\lambda x^2} \Rightarrow$$

$$\frac{\hbar^2}{m} \lambda - \frac{2\hbar^2}{m} \lambda^2 x^2 + \frac{\hbar^2}{2m} \lambda^2 x^2 = E \quad \Rightarrow$$

$$\frac{2\hbar^2}{m} \lambda^2 = \frac{\hbar^2}{2m} \lambda^2 \Rightarrow \lambda^2 = \frac{1}{4x_0^4} \Rightarrow \boxed{\lambda = \frac{1}{2x_0^2}}$$

(b) From eq. above

$$E = \frac{\hbar^2}{m} \lambda \Rightarrow \boxed{E = \frac{\hbar^2}{2m x_0^2}}$$

(c) For $-x_1 < x < 0$, $V(x) = 0$, so

$$\psi(x) = B \sin kx + C \cos kx$$

At $x=0$, $\psi'(x) = 0$ since $\frac{d\psi}{dx} = -2\lambda x e^{-\lambda x^2}$ for $x \geq 0$.

Therefore, $B = 0$. From continuity at $x=0$, $\boxed{C = A}$

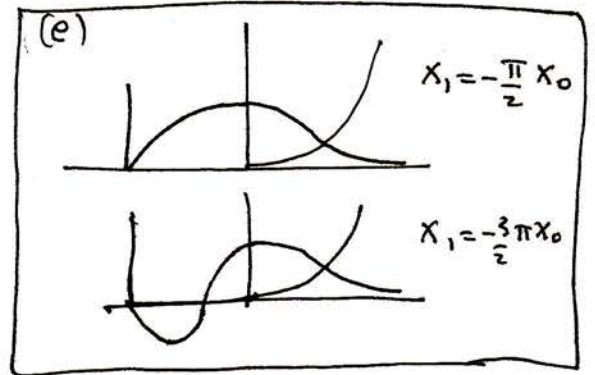
$$\text{The energy is } E = \frac{\hbar^2 k^2}{2m} = \frac{\hbar^2}{2m x_0^2} \Rightarrow \boxed{k = \frac{1}{x_0}} \Rightarrow$$

$$\boxed{\psi(x) = A \cos\left(\frac{x}{x_0}\right)}$$

(d) At x_1 , $V = \infty \Rightarrow \psi(x_1) = 0 \Rightarrow \frac{x_1}{x_0} = -\frac{\pi}{2}, -\frac{3}{2}\pi, \dots$

$$\Rightarrow \boxed{x_1 = -\frac{\pi}{2} x_0}, \quad \boxed{x_1 = -\frac{3}{2} \pi x_0} \quad \text{are 2 possible values.}$$

(e) See graph above



Problem 3

$$\psi(x) = Bx e^{-x/2x_0}, \quad x_0 = 2 \text{ \AA}$$

$$(a) \quad 1 = \int_0^{\infty} dx B^2 x^2 e^{-x/x_0} = B^2 \frac{2!}{\frac{1}{x_0^3}} = 2x_0^3 B^2 \Rightarrow B = \left(\frac{1}{2x_0^3}\right)^{1/2}$$

$$(b) \quad \langle x \rangle = \frac{\int_0^{\infty} dx x^3 e^{-x/x_0}}{\int_0^{\infty} dx x^2 e^{-x/x_0}} = \frac{3! x_0^4}{2! x_0^3} = \boxed{3x_0 = 6 \text{ \AA}}$$

$$(c) \quad \langle x^2 \rangle = \frac{\int_0^{\infty} dx x^4 e^{-x/x_0}}{\int_0^{\infty} dx x^2 e^{-x/x_0}} = \frac{4! x_0^5}{2! x_0^3} = 12x_0^2 \Rightarrow$$

$$\Delta x = \sqrt{12 - 9} x_0 = \boxed{\sqrt{3} x_0 = 1.732 x_0 = 3.46 \text{ \AA}}$$

$$(d) \quad p\psi = \frac{\hbar}{i} \frac{d}{dx} Bx e^{-x/2x_0} = \frac{\hbar}{i} B \left(e^{-x/2x_0} - \frac{x}{2x_0} e^{-x/2x_0} \right)$$

$$\langle p \rangle = \int dx \psi^* p\psi = \frac{\hbar}{i} B^2 \int_0^{\infty} dx \left(x e^{-x/x_0} - \frac{x^2}{2x_0} e^{-x/x_0} \right) =$$

$$= \frac{\hbar}{i} B^2 \left(1! x_0^2 - \frac{2!}{2} \frac{x_0^3}{x_0} \right) = \boxed{0}$$

$$(e) \quad \langle p^2 \rangle = \left\langle -\frac{\hbar^2}{2m} \frac{d^2}{dx^2} \right\rangle ; \quad \frac{d^2}{dx^2} x e^{-x/2x_0} = \frac{-2}{2x_0} e^{-x/2x_0} + \frac{x}{4x_0^2} e^{-x/2x_0}$$

$$\langle p^2 \rangle = \frac{\hbar^2 \int_0^{\infty} dx \left(\frac{x}{x_0} - \frac{x^2}{4x_0^2} \right) e^{-x/x_0}}{\int_0^{\infty} dx x^2 e^{-x/x_0}} = \frac{\hbar^2 \left(1! x_0 - \frac{2!}{4} x_0 \right)}{2! x_0^3} = \frac{\hbar^2}{4x_0^2}$$

$$\Rightarrow \Delta p = \sqrt{\langle p^2 \rangle - \langle p \rangle^2} = \frac{\hbar}{2x_0} ; \quad \boxed{\Delta p \Delta x = \frac{\hbar}{2x_0} \cdot \sqrt{3} x_0 = \frac{\sqrt{3}}{2} \hbar \sim \hbar \cdot 0.4.}$$

$$(f) \quad \langle E_{kin} \rangle = \langle p^2 \rangle / 2m = \boxed{\frac{\hbar^2}{8m_e x_0^2} = 0.238 \text{ eV}}$$

Problem 4

(a) $\Psi(x) = A \sin kx + B \cos kx$

Since $\Psi(x=0) = 0 \Rightarrow B = 0 \Rightarrow$

$$\boxed{\Psi(x) = A \sin kx}, \quad E = \frac{\hbar^2 k^2}{2m} \Rightarrow k = \left(\frac{2mE}{\hbar^2} \right)^{1/2}$$

Since $E = 2 \text{ eV} \Rightarrow \boxed{k = 0.7245 \text{ \AA}^{-1}}$

(b) $\boxed{\Psi(x) = C e^{-\alpha x}}, \quad \alpha = \sqrt{\frac{2m}{\hbar^2} (V_0 - E)}$

$\Rightarrow \alpha = \sqrt{\frac{2}{7.62} (6 - 2)} \text{ \AA}^{-1} \Rightarrow \boxed{\alpha = 1.025 \text{ \AA}^{-1}}$

(c) By continuity of Ψ and Ψ' ,

$$A \sin(kL) = C e^{-\alpha L} \Rightarrow \frac{1}{k} \tan(kL) = -\frac{1}{\alpha} \Rightarrow$$

$$kA \cos(kL) = -\alpha C e^{-\alpha L}$$

$\Rightarrow \tan(kL) = -\frac{k}{\alpha} = -0.7068 \Rightarrow kL = 2.526 \Rightarrow \boxed{L = 3.487 \text{ \AA}}$

(d) $\boxed{E_1 = \frac{\hbar^2 \pi^2}{2mL^2} = 3.09 \text{ eV}}$

(e) $2 \text{ eV} = \frac{\hbar^2 \pi^2}{2mL_{\text{eff}}^2} \Rightarrow \boxed{L_{\text{eff}} = 4.33 \text{ \AA}}$

(f) The "penetration" of the wavefunction in the "forbidden region" where $V_0 > E$ is $\sim \alpha^{-1} = \frac{1}{1.025} \text{ \AA} = 0.976 \text{ \AA}$.

If we add that to $L = 3.487 \text{ \AA}$, we get 4.46 \AA . That is pretty close to the effective length $L_{\text{eff}} = 4.33 \text{ \AA}$.

Problem 5

(a) ionization energy $\frac{E_0 Z^2}{n^2} = I$. $\lambda = 80 \text{ \AA}$, $\frac{hc}{\lambda} = 155 \text{ eV}$

$$E_{\text{ion}} = \frac{hc}{\lambda} - I \Rightarrow 32.6 \text{ eV} = 155 \text{ eV} - E_0 \frac{Z^2}{n^2} \Rightarrow$$

$$\frac{Z^2}{n^2} = \frac{155 - 32.6}{13.6} = 9 \Rightarrow \boxed{\frac{Z}{n} = 3}$$

$$\Rightarrow \boxed{n=1, Z=3 ; n=2, Z=6 ; n=3, Z=9 ; \dots}$$

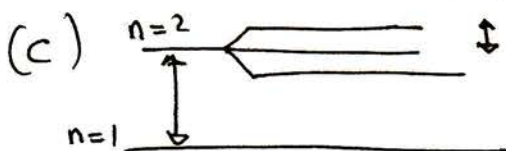
(b) Since a magnetic field gives 3-fold splitting $\Rightarrow l=1$.

The smallest n is $n=2 \Rightarrow \boxed{Z=6}$

Quantum numbers are $n=2, l=1, m_l=1, 0, -1$.

Wavefunctions:

$$\begin{aligned} \Psi(r, \theta, \phi) &= C e^{-3r/a_0} \frac{r}{a_0} \sin \theta e^{i m_l \phi}, m_l = \pm 1 \\ &= C e^{-3r/a_0} \left(1 - \frac{3r}{a_0}\right) \cos \theta, m_l = 0 \end{aligned}$$



$$\Delta E = E_0 \frac{Z^2}{n^2} \left(\frac{3}{4}\right) \pm \mu_B B = 367.2 \text{ eV} \pm \begin{matrix} 0.16 \text{ eV} \\ 0 \end{matrix}$$

$$\Delta E = \frac{hc}{\lambda} \Rightarrow \boxed{\begin{aligned} \lambda_1 &= 33.76 \text{ \AA} \\ \lambda_2 &= 33.77 \text{ \AA} \\ \lambda_3 &= 33.78 \text{ \AA} \end{aligned}}$$

(d) Spin-orbit splitting $\propto Z^4$, according to qu 4, prob 3.

For $Z=1$, it is $4.5 \times 10^{-5} \text{ eV}$. $Z^4 = 6^4 = 1296$

\Rightarrow spin-orbit splitting here is $\sim \boxed{5.8 \times 10^{-2} \text{ eV}}$

Problem 6

$$kT_E = \frac{1}{2} \hbar \omega \Rightarrow T_E = \frac{0.15 \text{ eV}}{\frac{1}{11,600} \text{ eV}} \Rightarrow \boxed{T_E = 1740^\circ \text{K}}$$

$$(b) \langle E \rangle = \frac{\frac{1}{2} \hbar \omega}{2} + \frac{\frac{1}{2} \hbar \omega}{e^{\frac{1}{2} \hbar \omega / kT} - 1}$$

$$kT \gg \frac{1}{2} \hbar \omega, \quad \langle E \rangle = kT \Rightarrow \boxed{\langle E \rangle = 86.2 \text{ eV}}$$

$$(c) \text{ at } T=0, \quad \boxed{E = \frac{1}{2} \hbar \omega = 0.075 \text{ eV}}$$

$$(d) \langle E \rangle = \frac{\frac{1}{2} \hbar \omega}{2} + \frac{\frac{1}{2} \hbar \omega}{e^1 - 1} = \frac{1}{2} \hbar \omega (0.5 + 0.582) = 1.082 \frac{1}{2} \hbar \omega$$
$$\Rightarrow \boxed{\langle E \rangle = 0.162 \text{ eV}}$$

(e) Boltzmann factor is $e^{-E/kT}$

\Rightarrow ratio of probabilities is:

$$\frac{P(E_2)}{P(E_1)} = e^{-(E_2 - E_1)/kT} = e^{-0.3 \text{ eV} / 0.15 \text{ eV}} = e^{-2} = 0.1353$$

\Rightarrow if there are 10,000 bullets with energy $E_1 = 1.575 \text{ eV}$,

$\boxed{\text{there are } 1353 \text{ bullets with energy } E_2 = 1.875 \text{ eV}}$

Problem 7

$$(a) E_1 = \frac{\hbar^2 \pi^2}{2mL^2} \Rightarrow \frac{1}{m_p L_1^2} = \frac{1}{m_e L_2^2} \Rightarrow$$

$$\Rightarrow \frac{L_2}{L_1} = \left(\frac{m_p}{m_e}\right)^{1/2} = \left(\frac{938.3}{0.511}\right)^{1/2} \Rightarrow \boxed{\frac{L_2}{L_1} = 42.9}$$

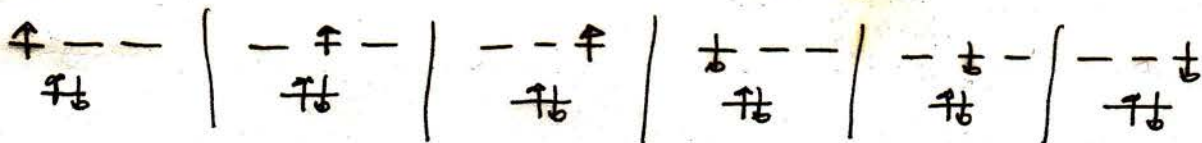
$$(b) T = e^{-2\sqrt{\frac{2m_e}{\hbar^2}(V_0-E)}} = e^{-2\sqrt{\frac{9}{3.81}}} = e^{-3.07} = 0.0462$$

\Rightarrow 46 electrons tunnel for every 1000 incident

(c) Lowest state: E_{111} , next: $E_{211} = E_{121} = E_{112}$

Each state can accommodate 2 electrons of opposite spin.

Different states:



There are 6 different states.