


Quiz 2 Solutions

2D - Dr. Sutterley

1.

Before



 $K_e = 1 \text{ MeV}$

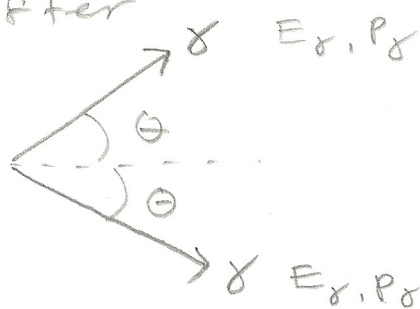
 $P = 1.422 \frac{\text{MeV}}{c}$



 $K_p = 0$

 $P = 0$

After



A.

Conservation of energy

$$E_{\text{total-before}} = E_{\text{total-after}}$$

$$KE_e + m_e c^2 + m_e c^2 = E_\gamma + E_\gamma$$

for the moving electron for the positron @ rest

or you could write

$$\sqrt{(P_e c)^2 + (m_e c^2)^2} + m_e c^2 = 2E_\gamma$$

$$\Rightarrow KE + 2m_e c^2 = 2E_\gamma$$

cons. of momentum

x-direction

$$P_e = 2p_\gamma \cos \theta$$

y-direction

$$0 = p_\gamma \sin \theta - p_\gamma \sin \theta \rightarrow \text{no information}$$

B. Since

$$KE + 2m_e c^2 = 2E_\gamma$$

$$E_\gamma = \frac{KE}{2} + m_e c^2 = \frac{1 \text{ MeV}}{2} + 0.511 \text{ MeV} = \boxed{1.011 \text{ MeV}}$$

C. using momentum cons.

$$P_e = 2p_\gamma \cos \theta$$

and $p_\gamma = \frac{E_\gamma}{c}$ because $E_\gamma^2 = (p_\gamma c)^2 + (m_e c^2)^2$ ^{0 for a photon}

$$E_\gamma = p_\gamma c$$

$$\text{so } \cos \theta = \frac{P_e}{2p_\gamma} = \frac{1.422 \text{ MeV}/c}{2(1.011 \text{ MeV}/c)} = 0.703$$

$$\Rightarrow \boxed{\theta = 45.3^\circ}$$

Answer in Radians is ok too...

$$2. \lambda = 500 \text{ nm}, V_s = 0.45 \text{ V}$$

$$A. \text{ Since } V_s e = KE_{\text{max}} = hf - \phi$$

$$KE_{\text{max}} = (0.45 \text{ V})(1.602 \times 10^{-19} \text{ C})(1 \text{ eV} / 1.602 \times 10^{-19} \text{ J})$$

$$\boxed{KE_{\text{max}} = 0.45 \text{ eV}}$$

$$B. \phi = hf - KE_{\text{max}} = \frac{hc}{\lambda} - KE_{\text{max}} = \frac{1240 \text{ eV} \cdot \text{nm}}{500 \text{ nm}} - 0.45 \text{ eV}$$

$$\boxed{\phi = 2.03 \text{ eV}}$$

C. λ_{cutoff} corresponds to the minimum energy the light must have to break out an electron w/ zero kinetic energy.

$$KE_{\text{max}} \rightarrow 0 = hf_{\text{cut}} - \phi \Rightarrow \phi = \frac{hc}{\lambda_{\text{cutoff}}}$$

$$\lambda_{\text{cutoff}} = \frac{hc}{\phi} = \frac{1240 \text{ eV} \cdot \text{nm}}{2.03 \text{ eV}} = \boxed{611 \text{ nm}}$$

3. Binding Energy is the difference in the mass between all the individual nucleons and the atomic mass.

$$B.E = \Delta m = [2m_p + 2m_n] - m_{\text{He}} =$$

$$[2(1.007276 \text{ u}) + 2(1.008665 \text{ u})] - 4.002602 \text{ u} =$$

$$= 0.02928 \text{ u} (931.49 \text{ MeV/u})$$

$$\boxed{BE = 27.27 \text{ MeV}}$$

$$4. \lambda_0 = 0.03 \text{ nm}$$

$$A. \Delta\lambda = \lambda' - \lambda_0 = \frac{h}{m_e c} (1 - \cos\theta) =$$

$$\lambda' = (0.00243 \text{ nm})(1 - \cos 90^\circ) + 0.03 \text{ nm}$$

$$\boxed{\lambda' = 0.03243 \text{ nm}}$$

B. Conservation of energy

$$\underbrace{E_{\gamma_0} + m_e c^2}_{\text{Before}} = \underbrace{KE_e + m_e c^2 + E_{\gamma'}}_{\text{after}}$$

$$\text{so } KE_e = E_{\gamma_0} - E_{\gamma'} = \frac{hc}{\lambda_0} - \frac{hc}{\lambda'} = (1240 \text{ nm}\cdot\text{eV}) \left[\frac{1}{0.03 \text{ nm}} - \frac{1}{0.03243 \text{ nm}} \right]$$

$$\boxed{KE = 3097 \text{ eV}}$$

5. B.