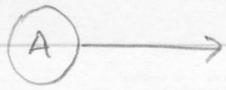


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$$m_A = 900 \text{ kg}, u_A = 0.85c$$



$$m_B = 140 \text{ kg}$$



completely inelastic collision

What is the velocity and mass of the composite object?

Unlike classical momentum, relativistic momentum

is conserved in inelastic collisions. So, we have:

$$\text{Energy conservation: } \gamma_A m_A c^2 + \gamma_B m_B c^2 = \gamma_C m_C c^2 \quad ①$$

$$\text{Momentum conservation: } \gamma_A m_A u_A = \gamma_C m_C u_C \quad ②$$

The unknowns here are m_C and u_C .

Use your favorite method to solve for these unknowns:

$$(a) \text{ Dividing } \frac{②}{①} \text{ yields } \frac{u_C}{C} = \frac{\gamma_A m_A u_A}{\gamma_A m_A c^2 + \gamma_B m_B c^2} = 0.46$$

$$\Rightarrow \boxed{u_C = 0.46c}$$

(b) From conservation of momentum (②)

$$m_C = \frac{\gamma_A m_A u_A}{\gamma_C u_C} = \boxed{2.81 \times 10^3 \text{ kg}}$$

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Going back to your 2A material, gravitational potential energy is given by:

$$U = -\frac{GMm}{r}$$

"Removing a piece from the surface" means moving m from $r = R_g$ to $r = \infty$

$$\begin{aligned}\Rightarrow \Delta U &= U(r=\infty) - U(r=R_g) \\ &= \frac{GMm}{R_g}\end{aligned}$$

They tell us that this is equal to the rest mass energy:

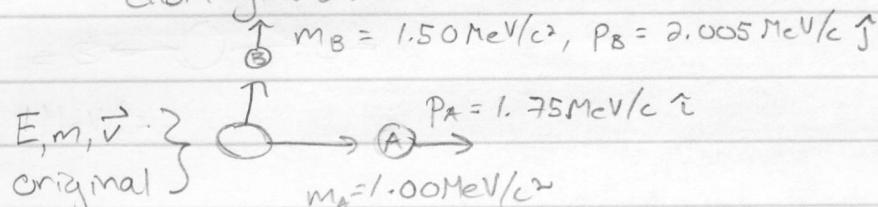
$$\begin{aligned}G\frac{Mm}{R_g} &= mc^2 \\ \Rightarrow R_g &= \frac{G\frac{Mm}{mc^2}}{c^2} = \frac{GM}{c^2}\end{aligned}$$

You can look up M = mass of the sun and

G = gravitational constant to get $R_g = [1.47 \times 10^3 \text{ m}]$

29 p.1

disintegration



What is the mass m and speed $|\vec{v}|$ of the original?

Again, here we simply use conservation of relativistic energy and momentum $\Rightarrow E = \sqrt{p_A^2 c^2 + m_A^2 c^4} + \sqrt{p_B^2 c^2 + m_B^2 c^4}$

Notice that p_A and p_B are orthogonal so $p_x = p_A$ and $p_y = p_B$

$$\Rightarrow p = \sqrt{p_A^2 + p_B^2}. \text{ To find the mass, use } E^2 = p^2 c^2 + m^2 c^4.$$

$$\Rightarrow m = [3.65 \text{ MeV}/c^2]. \text{ Total energy: } E = \gamma mc^2 \Rightarrow v = [0.589c]$$

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What is the peak wavelength for emission from a blackbody at 35°C?

Wien's displacement law says

$$\lambda_{\max} T = 2.898 \times 10^{-3} \text{ m K}$$

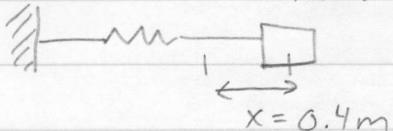
To use this we should convert 35°C to K.

$$35^\circ\text{C} = 308.15 \text{ K}$$

$$\lambda_{\max} = \frac{2.898 \times 10^{-3} \text{ m K}}{308.15 \text{ K}} = [9.405 \mu\text{m}]$$

#3 p. 1

$$k = 25 \text{ N/m} \quad m = 2 \text{ kg}$$



What is the total energy and frequency of oscillations classically?

$E = K + U$ is conserved, so we are free to calculate it at any time. The easiest time is right before it is released $\Rightarrow K = 0$

$$\Rightarrow E = U = \frac{1}{2} k x^2 = \frac{1}{2} (25 \text{ N/m}) (0.4 \text{ m})^2 = [2 \text{ J}]$$

$$\text{The frequency of oscillation is given by } \omega = \sqrt{\frac{k}{m}} = \sqrt{\frac{25 \text{ N/m}}{2 \text{ kg}}} = 3.53 \frac{\text{rad}}{\text{s}}$$

$$\text{or } f = \frac{\omega}{2\pi} = \frac{3.53 \text{ rad/s}}{2\pi} = [0.563 \text{ Hz}]$$

(b) Assume energy is quantized, find n.

$$E = nhf \Rightarrow n = \frac{E}{hf} = \frac{2 \text{ J}}{(6.626 \times 10^{-34} \frac{\text{J m}^2 \text{kg}}{\text{s}})(0.563 \text{ Hz})} = \overline{5.36 \times 10^{33}}$$

$$(c) E = hf = (6.626 \times 10^{-34} \frac{\text{J m}^2 \text{kg}}{\text{s}})(0.563 \text{ Hz}) = [3.73 \times 10^{-34} \text{ J}]$$

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Calculate the energy of a photon (in eV):

$$(a) f = 5 \times 10^{14} \text{ Hz}, E = hf = (4.136 \times 10^{-15} \text{ eV s})(5 \times 10^{14} \text{ Hz}) = [2.068 \text{ eV}]$$

$$(b) f = 10 \times 10^9 \text{ Hz}, E = (4.136 \times 10^{-15} \text{ eV})(10 \times 10^9 \text{ Hz}) = [4.14 \times 10^{-5} \text{ eV}]$$

$$(c) f = 30 \times 10^6 \text{ Hz}, E = [1.24 \times 10^{-7} \text{ eV}]$$

#10 p. 1

$$P = 100 \times 10^3 \text{ W}$$

$$f = 94 \times 10^6 \text{ Hz}$$

The energy of each photon is $E = hf = 6.23 \times 10^{-26} \text{ J}$

The number of photons per second is

$$\frac{\#}{s} = \frac{P}{E} = \frac{100 \times 10^3 \text{ J/s}}{6.23 \times 10^{-26} \text{ J}} = [1.61 \times 10^{30} / \text{s}]$$

#14 p. 1

$$K_{\max} = hf - \varphi = \frac{hc}{\lambda} - \varphi = \frac{(4.136 \times 10^{-15} \text{ eV s})(3 \times 10^8 \text{ m/s})}{350 \times 10^{-9} \text{ m}} - 2.24 \text{ eV}$$

$$= [1.31 \text{ eV}]$$

(b) The cutoff wavelength is the maximum wavelength

(\Rightarrow minimum energy) that can cause an e^- to

escape from the surface. We therefore assume $K = 0$.

$$0 = hf - \varphi = \frac{hc}{\lambda} - \varphi \Rightarrow \lambda_c = \frac{hc}{\varphi} = \frac{(4.136 \times 10^{-15} \text{ eV s})(3 \times 10^8 \text{ m/s})}{(2.24 \text{ eV})}$$

$$= [5.54 \times 10^{-7} \text{ m}]$$

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$$\phi = 4.2 \text{ eV}$$

This is the same as the last problem

$$\lambda_c = \frac{hc}{\phi} = \boxed{2.95 \times 10^{-7} \text{ m}} \Rightarrow f_c = \frac{c}{\lambda_c} = \boxed{1.02 \times 10^{15} \text{ Hz}}$$

The stopping potential is the potential necessary to achieve no current flow in the photo-electric experiment:

$$eV_s = K = hf - \phi = \frac{hc}{\lambda} - \phi$$

$$\Rightarrow V_s = \frac{hc}{\lambda e} - \frac{\phi}{e} = \frac{(1240 \text{ eV nm})}{200 \text{ nm (e)}} - \frac{4.2 \text{ eV}}{e} = \boxed{2.0 \text{ V}}$$