

6-43. (a) For $x > 0$, $\hbar^2 k_2^2 / 2m + V_0 = E = \hbar^2 k_1^2 / 2m = 2V_0$

So, $k_2 = (2mV_0)^{1/2} / \hbar$. Because $k_1 = (4mV_0)^{1/2} / \hbar$, then $k_2 = k_1 / \sqrt{2}$

(b) $R = (k_1 - k_2)^2 / (k_1 + k_2)^2$ (Equation 6-68)

$= (1 - 1/\sqrt{2})^2 / (1 + 1/\sqrt{2})^2 = 0.0294$, or 2.94% of the incident particles

are

reflected.

(c) $T = 1 - R = 1 - 0.0294 = 0.971$

(d) 97.1% of the particles, or $0.971 \times 10^6 = 9.71 \times 10^5$, continue past the step in the $+x$

direction. Classically, 100% would continue on.

6-44. (a) For $x > 0$, $\hbar^2 k_2^2 / 2m - V_0 = E = \hbar^2 k_1^2 / 2m = 2V_0$

So, $k_2 = (6mV_0)^{1/2} / \hbar$. Because $k_1 = (4mV_0)^{1/2} / \hbar$, then $k_2 = \sqrt{3/2} k_1$

(b) $R = (k_1 - k_2)^2 / (k_1 + k_2)^2$

$R = (k_1 - k_2)^2 / (k_1 + k_2)^2 = (1 - \sqrt{3/2})^2 / (1 + \sqrt{3/2})^2 = 0.0102$

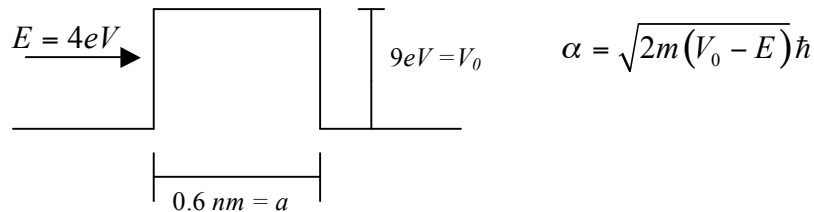
Or 1.02% are reflected at $x = 0$.

(c) $T = 1 - R = 1 - 0.0102 = 0.99$

(d) 99% of the particles, or $0.99 \times 10^6 = 9.9 \times 10^5$, continue in the $+x$ direction.

Classically, 100% would continue on.

6-45. (a)



and $\alpha a = 0.6 \text{ nm} \times 11.46 \text{ nm}^{-1} = 6.87$

Since αa is not $\ll 1$, use Equation 6-75:

The transmitted fraction

$$T = \left[1 + \frac{\sinh^2 \alpha a}{4(E/V_0)(1 - E/V_0)} \right]^{-1} = \left[1 + \left(\frac{81}{80} \right) \sinh^2 (6.87) \right]^{-1}$$

Recall that $\sinh x = (e^x - e^{-x})/2$,

$$T = \left[1 + \frac{81}{80} \left(\frac{e^{6.87} - e^{-6.87}}{2} \right)^2 \right]^{-1} = 4.3 \times 10^{-6} \text{ is the transmitted}$$

fraction.

(b) Noting that the size of T is controlled by αa through the $\sinh^2 \alpha a$ and increasing T

implies increasing E . Trying a few values, selecting $E = 4.5 \text{ eV}$ yields

$$T = 8.7 \times 10^{-6}$$

or approximately twice the value in part (a).

6-48. Using Equation 6-76,

$$T \approx 16 \frac{E}{V_0} \left(1 - \frac{E}{V_0} \right) e^{-2\alpha a} \text{ where } E = 2.0 \text{ eV}, V_0 = 6.5 \text{ eV}, \text{ and } a = 0.5 \text{ nm}.$$

$$T \approx 16 \left(\frac{2.0}{6.5} \right) \left(1 - \frac{2.0}{6.5} \right) e^{-2(10.87)(0.5)} \approx 6.5 \times 10^{-5} \quad (\text{Equation 6-75 yields}$$

$$T = 6.6 \times 10^{-5}.)$$

$$6-49. \quad R = \frac{(k_1 - k_2)^2}{(k_1 + k_2)^2} \text{ and } T = 1 - R \quad (\text{Equations 6-68 and 6-70})$$

(a) For protons:

$$k_1 = \sqrt{2mc^2 E} / \hbar c = \sqrt{2(938 \text{ MeV})(40 \text{ MeV})} / 197.3 \text{ MeV} \cdot \text{fm} = 1.388$$

$$k_2 = \sqrt{2mc^2 (E - V_0)} / \hbar c = \sqrt{2(938 \text{ MeV})(10 \text{ MeV})} / 197.3 \text{ MeV} \cdot \text{fm} = 0.694$$

$$R = \left(\frac{1.388 - 0.694}{1.388 + 0.694} \right)^2 = \left(\frac{0.694}{2.082} \right)^2 = 0.111 \quad \text{And } T = 1 - R = 0.889$$

(b) For electrons:

$$k_1 = 1.388 \left(\frac{0.511}{938} \right)^{1/2} = 0.0324 \quad k_2 = 0.694 \left(\frac{0.511}{938} \right)^{1/2} = 0.0162$$

$$R = \left(\frac{0.0324 - 0.0162}{0.0324 + 0.0162} \right)^2 = 0.111 \quad \text{And } T = 1 - R = 0.889$$

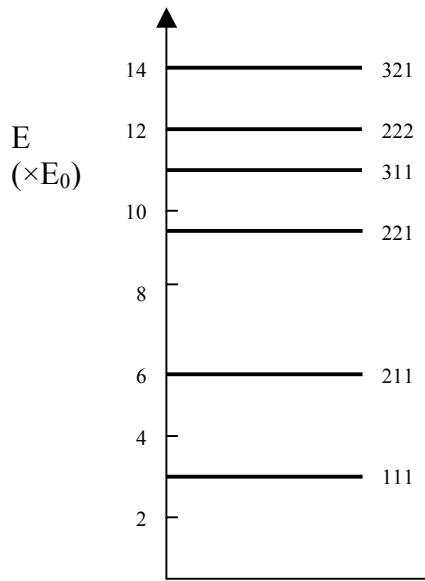
No, the mass of the particle is not a factor. (We might have noticed that \sqrt{m} could be canceled from each term.)

7-1. $E_{n_1 n_2 n_3} = \frac{\hbar^2 \pi^2}{2mL^2} (n_1^2 + n_2^2 + n_3^2)$ (Equation 7-4)

$$E_{311} = \frac{\hbar^2 \pi^2}{2mL^2} (3^2 + 1^2 + 1^2) = 11E_0 \quad \text{where } E_0 = \frac{\hbar^2 \pi^2}{2mL^2}$$

$$E_{222} = E_0 (2^2 + 2^2 + 2^2) = 12E_0 \quad \text{and} \quad E_{321} = E_0 (3^2 + 2^2 + 1^2) = 14E_0$$

The 1st, 2nd, 3rd, and 5th excited states are degenerate.



$$7-2. \quad E_{n_1 n_2 n_3} = \frac{\hbar^2 \pi^2}{2m} \left(\frac{n_1^2}{L_1^2} + \frac{n_2^2}{L_2^2} + \frac{n_3^2}{L_3^2} \right) = \frac{\hbar^2 \pi^2}{2mL_1^2} \left(n_1^2 + \frac{n_2^2}{4} + \frac{n_3^2}{9} \right) \quad (\text{Equation 7-5})$$

$n_1 = n_2 = n_3 = 1$ is the lowest energy level.

$$E_{111} = E_0 \left(1 + 1/4 + 1/9 \right) = 1.361E_0 \quad \text{where } E_0 = \frac{\hbar^2 \pi^2}{2mL_1^2}$$

The next nine levels are, increasing order,

(Problem 7-2 continued)

n_1	n_2	n_3	$E (\times E_0)$
1	1	2	1.694
1	2	1	2.111
1	1	3	2.250
1	2	2	2.444
1	2	3	3.000
1	1	4	3.028
1	3	1	3.360
1	3	2	3.472
1	2	4	3.778

7-3. (a) $\psi_{n_1 n_2 n_3}(x, y, z) = A \cos \frac{n_1 \pi x}{L} \sin \frac{n_2 \pi y}{L} \sin \frac{n_3 \pi z}{L}$

(b) They are identical. The location of the coordinate origin does not affect the energy

level structure.

7-4. $\psi_{111}(x, y, z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{2L_1} \sin \frac{\pi z}{3L_1}$

$$\psi_{112}(x, y, z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{2L_1} \sin \frac{2\pi z}{3L_1}$$

$$\psi_{121}(x, y, z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{L_1} \sin \frac{\pi z}{3L_1}$$

$$\psi_{122}(x, y, z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{L_1} \sin \frac{2\pi z}{3L_1}$$

$$\psi_{113}(x, y, z) = A \sin \frac{\pi x}{L_1} \sin \frac{\pi y}{2L_1} \sin \frac{\pi z}{L_1}$$

7-5.

$$E_{n_1 n_2 n_3} = \frac{\hbar^2 \pi^2}{2m} \left(\frac{n_1^2}{L_1^2} + \frac{n_2^2}{(2L_1)^2} + \frac{n_3^2}{(4L_1)^2} \right) = \frac{\hbar^2 \pi^2}{2mL_1^2} \left(n_1^2 + \frac{n_2^2}{4} + \frac{n_3^2}{16} \right) \quad (\text{from Equation 7-5})$$

$$E_0 = \left(n_1^2 + \frac{n_2^2}{4} + \frac{n_3^2}{16} \right) \quad \text{where } E_0 = \frac{\hbar^2 \pi^2}{2mL_1^2}$$

(Problem 7-5 continued)

(a)

n_1	n_2	n_3	$E (\times E_0)$
1	1	1	1.313
1	1	2	1.500
1	1	3	1.813
1	2	1	2.063
1	1	4	2.250
1	2	2	2.250
1	2	3	2.563
1	1	5	2.813
1	2	4	3.000
1	1	6	3.500

(b) 1,1,4 and 1,2,2

$$7-7. \quad E_0 = \frac{\hbar^2 \pi^2}{2mL^2} = \frac{(1.055 \times 10^{-34} \text{ J}\cdot\text{s})^2 \pi^2}{2(9.11 \times 10^{-31} \text{ kg})(0.10 \times 10^{-9} \text{ m})^2 (1.609 \times 10^{-19} \text{ J/eV})} = 37.68 \text{ eV}$$

$$E_{311} - E_{111} = \Delta E = 11E_0 - 3E_0 = 8E_0 = 301 \text{ eV}$$

(Problem 7-7 continued)

$$E_{222} - E_{111} = \Delta E = 12E_0 - 3E_0 = 9E_0 = 339 \text{ eV}$$

$$E_{321} - E_{111} = \Delta E = 14E_0 - 3E_0 = 11E_0 = 415eV$$

7-8. (a) Adapting Equation 7-3 to two dimensions (i.e., setting $k_3 = 0$), we have

$$\psi_{n_1 n_2} = A \sin \frac{n_1 \pi x}{L} \sin \frac{n_2 \pi y}{L}$$

(b) From Equation 7-5, $E_{n_1 n_2} = \frac{\hbar^2 \pi^2}{2mL^2} (n_1^2 + n_2^2)$

(c) The lowest energy degenerate states have quantum numbers $n_1 = 1, n_2 = 2$,
and $n_1 = 2,$

$$n_2 = 1.$$