

**Formulas:**

First law  $\Delta U = Q + W$  ;  $C_V = \left(\frac{\partial U}{\partial T}\right)_V$  ;  $H = U + PV$  ;  $C_P = \left(\frac{\partial H}{\partial T}\right)_P$

$k = 1.381 \times 10^{-23} \text{ J/K}$  ,  $N_A = 6.02 \times 10^{23}$  ,  $R = 8.31 \text{ J/(mol}^\circ\text{K)}$

Ideal gas :  $PV = NkT$   $U = N \frac{f}{2} kT$   $W = -P\Delta V$  ,  $W = -\int PdV$  ;  $C_P = C_V + Nk$

adiabatic process:  $PV^\gamma = \text{const.}$  ,  $\gamma = (f + 2)/f$

$S = k \ln \Omega$  ;  $\frac{1}{T} = \left(\frac{\partial S}{\partial U}\right)_V$  ;  $dS = \frac{Q}{T} = \frac{C_V dT}{T}$  (constant volume)

Stirling's approximation :  $n! = n^n e^{-n} \sqrt{2\pi n}$

Ideal gas (monoatomic) :  $\Omega_N = C_N V^N U^{3N/2} / N!$

Einstein solid :  $\Omega(N, q) = \frac{(q + N - 1)!}{q!(N - 1)!} \sim \left(\frac{eq}{N}\right)^N$  for  $q \gg N$

Two - state system :  $\Omega = \frac{N!}{N_\uparrow! N_\downarrow!}$

Paramagnetism :  $M = N\mu \tanh(\mu B/kT)$  ,  $U = -MB$  ,  $M = \mu(N_\uparrow - N_\downarrow)$

$dU = TdS - PdV$  ;  $Q = TdS$  ,  $W = -PdV$  (quasistatic)

$U(S, V)$  ;  $F(T, V) = U - TS$  ;  $H(S, P) = U + PV$  ;  $G(T, P) = U - TS + PV$

$C_V = T \left(\frac{\partial S}{\partial T}\right)_V$  ;  $C_P = T \left(\frac{\partial S}{\partial T}\right)_P$  ;  $\beta = \frac{1}{V} \left(\frac{\partial V}{\partial T}\right)_P$  ;  $\kappa_{T,S} = -\frac{1}{V} \left(\frac{\partial V}{\partial P}\right)_{T,S}$  ;  $\mu = -T \left(\frac{\partial S}{\partial N}\right)_{U,V} = \left(\frac{\partial U}{\partial N}\right)_{S,V}$

$\mu = -kT \ln\left(\frac{V}{N} \left(\frac{2\pi m kT}{h^2}\right)\right)$  Ideal gas (monoatomic)

Heat engines :  $e = \frac{W}{Q_h} \leq 1 - \frac{T_c}{T_h}$  ; Refrigerators :  $COP = \frac{Q_c}{W} \leq \frac{T_c}{T_h - T_c}$

Maxwell relations :  $\left(\frac{\partial T}{\partial V}\right)_S = -\left(\frac{\partial P}{\partial S}\right)_V$  ;  $\left(\frac{\partial T}{\partial P}\right)_S = \left(\frac{\partial V}{\partial S}\right)_P$  ;  $\left(\frac{\partial S}{\partial V}\right)_T = \left(\frac{\partial P}{\partial T}\right)_V$  ;  $\left(\frac{\partial S}{\partial P}\right)_T = -\left(\frac{\partial V}{\partial T}\right)_P$

Van der Waals :  $(P + a \frac{N^2}{V^2})(V - Nb) = NkT$  ; Virial :  $PV = NkT(1 + \frac{N}{V} B_2(T) + \frac{N^2}{V^2} B_3(T) + \dots)$

Boltzmann distribution :  $P(E_i) = C e^{-E_i/kT}$  ;  $C = 1 / \sum_i e^{-E_i/kT} \equiv 1/Z$  ;  $P(E) = C \Omega(E) e^{-E/kT}$

**Justify all your answers to all problems**