

**Formulas:**

Time dilation; Length contraction :  $\Delta t = \gamma \Delta t' \equiv \gamma \Delta t_p$  ;  $L = L_p / \gamma$  ;  $c = 3 \times 10^8 \text{ m/s}$

Lorentz transformation :  $x' = \gamma(x - vt)$  ;  $y' = y$  ;  $z' = z$  ;  $t' = \gamma(t - vx/c^2)$  ; inverse :  $v \rightarrow -v$

Spacetime interval :  $(\Delta s)^2 = (c\Delta t)^2 - [\Delta x^2 + \Delta y^2 + \Delta z^2]$

Velocity transformation :  $u_x' = \frac{u_x - v}{1 - u_x v/c^2}$  ;  $u_y' = \frac{u_y}{\gamma(1 - u_x v/c^2)}$  ; inverse :  $v \rightarrow -v$

Relativistic Doppler shift :  $f_{obs} = f_{source} \sqrt{1+v/c} / \sqrt{1-v/c}$  (approaching)

Momentum :  $\vec{p} = \gamma m \vec{u}$  ; Energy :  $E = \gamma mc^2$  ; Kinetic energy :  $K = (\gamma - 1)mc^2$

Rest energy :  $E_0 = mc^2$  ;  $E = \sqrt{p^2 c^2 + m^2 c^4}$

Electron :  $m_e = 0.511 \text{ MeV}/c^2$  Proton :  $m_p = 938.26 \text{ MeV}/c^2$  Neutron :  $m_n = 939.55 \text{ MeV}/c^2$

Atomic mass unit :  $1 u = 931.5 \text{ MeV}/c^2$  ; electron volt :  $1 \text{ eV} = 1.6 \times 10^{-19} \text{ J}$

Stefan's law :  $e_{tot} = \sigma T^4$  ,  $e_{tot}$  = power/unit area ;  $\sigma = 5.67 \times 10^{-8} \text{ W/m}^2 \text{K}^4$

$e_{tot} = cU/4$  ,  $U$  = energy density =  $\int_0^\infty u(\lambda, T) d\lambda$  ; Wien's law :  $\lambda_m T = \frac{hc}{4.96 k_B}$

Boltzmann distribution :  $P(E) = C e^{-E/(k_B T)}$

Planck's law :  $u_\lambda(\lambda, T) = N_\lambda(\lambda) \times \bar{E}(\lambda, T) = \frac{8\pi}{\lambda^4} \times \frac{hc/\lambda}{e^{hc/\lambda k_B T} - 1}$  ;  $N(f) = \frac{8\pi f^2}{c^3}$

Photons :  $E = hf = pc$  ;  $f = c/\lambda$  ;  $hc = 12,400 \text{ eV \AA}$  ;  $k_B = (1/11,600) \text{ eV/K}$

Photoelectric effect :  $eV_s = K_{max} = hf - \phi$  ,  $\phi$  = work function; Bragg equation :  $n\lambda = 2d \sin \theta$

Compton scattering :  $\lambda' - \lambda = \frac{h}{m_e c} (1 - \cos \theta)$  ;  $\frac{h}{m_e c} = 0.0243 \text{ \AA}$  ; Coulomb constant :  $ke^2 = 14.4 \text{ eV \AA}$

Force in electric and magnetic fields (Lorentz force) :  $\vec{F} = q\vec{E} + q\vec{v} \times \vec{B}$  ; Drag force :  $D = 6\pi\eta r v$

Rutherford scattering :  $\Delta n = \frac{C}{\sin^4(\phi/2)}$  ;  $\hbar c = 1,973 \text{ eV \AA}$

Hydrogen spectrum :  $\frac{1}{\lambda_{mn}} = R \left( \frac{1}{m^2} - \frac{1}{n^2} \right)$  ;  $R = 1.097 \times 10^7 \text{ m}^{-1} = \frac{1}{911.3 \text{ \AA}}$

Electrostatic force, energy :  $F = \frac{kq_1 q_2}{r^2}$  ;  $U = \frac{kq_1 q_2}{r}$  . Centripetal force :  $F_c = \frac{mv^2}{r}$

Bohr atom :  $E_n = -\frac{ke^2 Z}{2r_n} = -\frac{Z^2 E_0}{n^2}$  ;  $E_0 = \frac{ke^2}{2a_0} = 13.6 \text{ eV}$  ;  $K = \frac{m_e v^2}{2}$  ;  $U = -\frac{ke^2 Z}{r}$

$hf = E_i - E_f$  ;  $r_n = r_0 n^2$  ;  $r_0 = \frac{a_0}{Z}$  ;  $a_0 = \frac{\hbar^2}{m_e ke^2} = 0.529 \text{ \AA}$  ;  $L = m_e v r = n\hbar$  angular momentum

de Broglie :  $\lambda = \frac{h}{p}$  ;  $f = \frac{E}{h}$  ;  $\omega = 2\pi f$  ;  $k = \frac{2\pi}{\lambda}$  ;  $E = \hbar\omega$  ;  $p = \hbar k$  ;  $E = \frac{p^2}{2m}$

Wave packets :  $y(x, t) = \sum_j a_j \cos(k_j x - \omega_j t)$ , or  $y(x, t) = \int dk a(k) e^{i(kx - \omega(k)t)}$  ;  $\Delta k \Delta x \sim 1$  ;  $\Delta \omega \Delta t \sim 1$

group and phase velocity :  $v_g = \frac{d\omega}{dk}$  ;  $v_p = \frac{\omega}{k}$  ; Heisenberg :  $\Delta x \Delta p \sim \hbar$  ;  $\Delta t \Delta E \sim \hbar$

Schrodinger equation :  $-\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + U(x)\Psi(x, t) = i\hbar \frac{\partial \Psi}{\partial t}$  ;  $\Psi(x, t) = \psi(x) e^{-i\frac{E}{\hbar} t}$

Time – independent Schrodinger equation :  $-\frac{\hbar^2}{2m} \frac{\partial^2 \psi}{\partial x^2} + U(x)\psi(x) = E\psi(x)$  ;  $\int_{-\infty}^{\infty} dx \psi^* \psi = 1$

$\infty$  square well :  $\psi_n(x) = \sqrt{\frac{2}{L}} \sin\left(\frac{n\pi x}{L}\right)$  ;  $E_n = \frac{\pi^2 \hbar^2 n^2}{2mL^2}$  ;  $\frac{\hbar^2}{2m_e} = 3.81eVA^2$  (electron)

Harmonic oscillator :  $\Psi_n(x) = H_n(x)e^{-\frac{m\omega}{2\hbar}x^2}$  ;  $E_n = (n + \frac{1}{2})\hbar\omega$  ;  $E = \frac{p^2}{2m} + \frac{1}{2}m\omega^2 x^2 = \frac{1}{2}m\omega^2 A^2$  ;  $\Delta n = \pm 1$

**Justify all your answers to all problems**

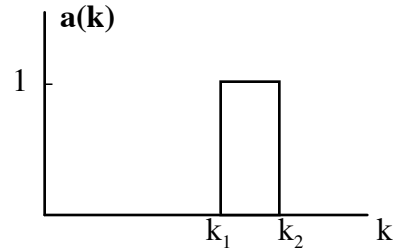
**Problem 1** (10 points)

A free electron (not subject to any forces) is described

by the wavepacket  $\psi(x,t) = \int dk a(k) e^{i(kx - \omega(k)t)}$

with  $a(k)$  given in the figure, with

$k_1 = 299A^{-1}$ ,  $k_2 = 301A^{-1}$



- (a) Estimate the uncertainty in the position of this electron, in A.
- (b) Find the speed of this electron, approximately.
- (c) Calculate  $\psi(x,t = 0)$  and make a graph of it versus  $x$ .

**Problem 2** (10 points)

A proton and a neutron attract each other through the strong nuclear force and form a bound state, called the deuteron. Assume the attractive potential can be described by the expression

$$U(x) = C \frac{x^2}{a^2}$$

as a function of the distance  $x$  between the proton and the neutron, where  $C$  and  $a$  are constants. Assume also the size of the deuteron is  $a$ , i.e.  $a$  is the (average) distance between the proton and neutron in the deuteron.

- (a) The size of the deuteron is experimentally found to be approximately 1 F (1F=10<sup>-5</sup> A), so  $a=10^{-5}$  A in the above formula. Estimate the value of the constant  $C$ .
- (b) If instead  $a$  had the value 0.529A, meaning the size of the deuteron would be the same as the size of a hydrogen atom,, what would the value of the constant  $C$  in the above formula be?

Hint: you may assume one of the nucleons is at rest, and minimize the energy of the other nucleon.

**Problem 3** (10 points)

An electron is in a box of length 6A in the lowest energy state (ground state).

- (a) Find its energy.
- (b) How much more likely is this electron to be found at the center of the box versus at distance 1A from one (any one) of the walls?
- (c) What is the largest wavelength photon that this electron can absorb in making a transition to another state? Give your answer in A. (Angstroms)

**Justify all your answers to all problems**