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4-14 (a) $r_n = \frac{n^2 \hbar^2}{m_e k e^2}$; where $n = 1, 2, 3, \dots$

$$r_n = n^2 \frac{(1.055 \times 10^{-34} \text{ Js})^2}{(9.11 \times 10^{-31} \text{ kg})(9.0 \times 10^9 \text{ Nm}^2/\text{C}^2)(1.6 \times 10^{-19} \text{ C})^2} = (0.0529 \text{ nm})n^2$$

For $n = 1$: $r_n = 0.0529 \text{ nm}$

For $n = 2$: $r_n = 0.2121 \text{ nm}$

For $n = 3$: $r_n = 0.4772 \text{ nm}$

(b) From Equation 4.26, $v = \left(\frac{ke^2}{m_e r}\right)^{1/2}$

$$v_1 = \left[\frac{(8.99 \times 10^9 \text{ Nm}^2/\text{C}^2)(1.60 \times 10^{-19} \text{ C})^2}{(9.11 \times 10^{-31} \text{ kg})(0.0529 \times 10^{-9} \text{ m})} \right]^{1/2} = 2.19 \times 10^6 \text{ m/s}$$

$$v_2 = 1.09 \times 10^6 \text{ m/s}$$

$$v_3 = 7.28 \times 10^5 \text{ m/s}$$

(c) As $c = 3.0 \times 10^8 \text{ m/s}$, $v \ll c$ and no relativistic correction is necessary.

4-17 $r = \frac{n^2 \hbar^2}{Z m_e k e^2} = \left(\frac{n^2}{Z}\right) \left(\frac{\hbar^2}{m_e k e^2}\right); n = 1$

$$r = \frac{1}{Z} \left[\frac{(1.055 \times 10^{-34} \text{ Js})^2}{(9.11 \times 10^{-31} \text{ kg})(9 \times 10^9 \text{ Nm}^2/\text{C}^2)(1.6 \times 10^{-19} \text{ C})^2} \right] = \frac{5.30 \times 10^{-11} \text{ m}}{Z}$$

(a) For He^+ , $Z = 2$, $r = \frac{5.30 \times 10^{-11} \text{ m}}{2} = 2.65 \times 10^{-11} \text{ m} = 0.0265 \text{ nm}$

(b) For Li^{2+} , $Z = 3$, $r = \frac{5.30 \times 10^{-11} \text{ m}}{3} = 1.77 \times 10^{-11} \text{ m} = 0.0177 \text{ nm}$

(c) For Be^{3+} , $Z = 4$, $r = \frac{5.30 \times 10^{-11} \text{ m}}{4} = 1.32 \times 10^{-11} \text{ m} = 0.0132 \text{ nm}$

4-22 $E = K + U = \frac{mv^2}{2} - \frac{ke^2}{r}$. But $\frac{mv^2}{2} = \left(\frac{1}{2}\right) \frac{ke^2}{r}$. Thus $E = \left(\frac{1}{2}\right) \left(\frac{-ke^2}{r}\right) = \frac{U}{2}$, so

$$U = 2E = 2(-13.6 \text{ eV}) = -27.2 \text{ eV} \text{ and } K = E - U = -13.6 \text{ eV} - (-27.2 \text{ eV}) = 13.6 \text{ eV}.$$

4-23 (a) $r_1 = (0.0529 \text{ nm})n^2 = 0.0529 \text{ nm}$ (when $n = 1$)

$$(b) \quad m_e v = m_e \left(\frac{ke^2}{m_e r} \right)^{1/2}$$

$$m_e = \left[\frac{(9.1 \times 10^{-31} \text{ kg})(9 \times 10^9 \text{ Nm}^2/\text{C}^2)}{5.29 \times 10^{-11} \text{ m}} \right]^{1/2} \times (1.6 \times 10^{-19} \text{ C})$$

$$M_e v = 1.99 \times 10^{-24} \text{ kg m/s}$$

$$(c) \quad L = m_e v r = (1.99 \times 10^{-24} \text{ kg m/s})(5.29 \times 10^{-11} \text{ m}), \quad L = 1.05 \times 10^{-34} (\text{kg m}^2/\text{s}) = \hbar$$

$$(d) \quad K = |E| = 13.6 \text{ eV}$$

$$(e) \quad U = -2K = -27.2 \text{ eV}$$

$$(f) \quad E = K + U = -13.6 \text{ eV}$$

4-25 (a) $\Delta E = hf = (13.6 \text{ eV}) \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$ or $f = (13.6 \text{ eV}) \left(\frac{\frac{1}{9} - \frac{1}{16}}{4.14 \times 10^{-15} \text{ eV s}} \right) = 1.60 \times 10^{14} \text{ Hz}$

(b) $T = \frac{2\pi r_n}{v}$ so $f_{\text{rev}} = \frac{1}{T} = \frac{v}{2\pi r_n}$. Using $v = \left(\frac{ke^2}{m_e r_n} \right)^{1/2}$, $f_{\text{rev}} = \left(\frac{ke^2}{mr_n} \right)^{1/2}$. For $n = 3$, $r_3 = (3)^2 a_0$ and

$$f_{\text{rev}} = \frac{(8.99 \times 10^9 \text{ Nm}^2/\text{C}^2)(1.60 \times 10^{-19} \text{ C})^2}{[(9.11 \times 10^{-31} \text{ kg})(9)(5.29 \times 10^{-11} \text{ m})]^{1/2}}$$

$$f_{\text{rev}} = 2.44 \times 10^{14} \text{ Hz } (n = 3)$$

$$f_{\text{rev}} = 1.03 \times 10^{14} \text{ Hz } (n = 4)$$

Thus the photon frequency is about halfway between the two frequencies of the revolution.

4-32 (a) $\mu_{\text{IH}} = \frac{m_e M}{m_e + M} = \frac{(9.109 390 \times 10^{-31} \text{ kg})(1.672 63 \times 10^{-27} \text{ kg})}{(9.109 390 \times 10^{-31} \text{ kg}) + (1.672 63 \times 10^{-27} \text{ kg})}$

$$= \frac{(9.109 390 \times 10^{-31} \text{ kg})(1.672 63 \times 10^{-27} \text{ kg})}{(0.000 910 939 0 \times 10^{-27} \text{ kg}) + (1.672 63 \times 10^{-27} \text{ kg})} = 9.104 431 \times 10^{-31} \text{ kg}$$

$$\frac{1}{\lambda} = \frac{\mu}{m_e} (k) \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right) = \left(\frac{9.104 431 6 \times 10^{-31} \text{ kg}}{9.109 390 \times 10^{-31} \text{ kg}} \right) \left(1.097 315 3 \times 10^7 \text{ m}^{-1} \right) \left(\frac{1}{2^2} - \frac{1}{3^2} \right)$$

$$\lambda_{\text{IH}} = 656.469 \text{ nm}$$

(b) Similarly, we find $\lambda_{\text{2H}} = 656.292 \text{ nm}$.

(c) $\lambda_{\text{3H}} = 656.232 \text{ nm}$

4-33 $r_n = \frac{n^2 a_0}{Z} = \frac{n^2 \hbar^2}{m k e^2 Z} = (1)^2 \frac{(1.05 \times 10^{-34} \text{ Js})^2}{(207 m_e)(8.99 \times 10^9 \text{ Nm}^2/\text{C})(1.60 \times 10^{-19} \text{ C})^2 (82)} = 3.1 \times 10^{-15} \text{ m}.$

Note that this means the muon grazes the nuclear surface, and so experiments with muonic atoms give information about the nuclear charge distribution.

$$E_n = -\frac{[(ke^2 Z^2)/(2a_0 n^2)]/ke^2 Z^2}{2\hbar^2 n^2/mk^2} = \frac{mk^2 e^4 Z^2}{2\hbar^2 n^2}$$

Using $m = 207 m_e$, $n = 1$, and $Z = 82$ yields $E_1 = 18.9 \text{ MeV}$. (Using the reduced mass makes no difference in the answers to three significant figures.)

5

- 5-2 The issue is: Can we use the simpler classical expression $p = (2mK)^{1/2}$ instead of the exact relativistic expression $p = \frac{K(1 + \frac{2mc^2}{K})^{1/2}}{c}$? As the relativistic expression reduces to $p = (2mK)^{1/2}$ for $K \ll 2mc^2$, we can use the classical expression whenever $K \ll 1 \text{ MeV}$ because mc^2 for the electron is 0.511 MeV .

(a) Here $50 \text{ eV} \ll 1 \text{ MeV}$, so $p = (2mK)^{1/2}$

$$\begin{aligned} \lambda &= \frac{h}{p} = \frac{h}{\left[(2) \left(\frac{0.511 \text{ MeV}}{c^2} \right) (50 \text{ eV}) \right]^{1/2}} = \frac{hc}{\left[(2)(0.511 \text{ MeV})(50 \text{ eV}) \right]^{1/2}} \\ &= \frac{1240 \text{ eV nm}}{\left[(2)(0.511 \times 10^6)(50)(\text{eV})^2 \right]^{1/2}} = 0.173 \text{ nm} \end{aligned}$$

(b) As $50 \text{ eV} \ll 1 \text{ MeV}$, $p = (2mK)^{1/2}$

$$\lambda = \frac{hc}{\left[(2) \left(\frac{0.511 \text{ MeV}}{c^2} \right) (50 \times 10^3 \text{ eV}) \right]^{1/2}} = 5.49 \times 10^{-3} \text{ nm}.$$

As this is clearly a worse approximation than in (a) to be on the safe side use the

relativistic expression for p : $p = K \frac{\left(1 + \frac{2mc^2}{K}\right)^{1/2}}{c}$ so

$$\begin{aligned} \lambda &= \frac{h}{p} = \frac{hc}{(K^2 + 2Kmc^2)^{1/2}} = \frac{1240 \text{ eV nm}}{\left[(50 \times 10^3)^2 + (2)(50 \times 10^3)(0.511 \times 10^6 \text{ eV}) \right]^{1/2}} \\ &= 5.36 \times 10^{-3} \text{ nm} = 0.00536 \text{ nm} \end{aligned}$$

5-3 $\lambda = \frac{h}{p} = \frac{h}{mv} = \frac{6.63 \times 10^{-34} \text{ Js}}{74 \text{ kg}} (5 \text{ m/s}) = 1.79 \times 10^{-36} \text{ m}$

5-7 A 10 MeV proton has $K = 10 \text{ MeV} \ll 2mc^2 = 1877 \text{ MeV}$ so we can use the classical expression $p = (2mK)^{1/2}$. (See Problem 5-2)

$$\lambda = \frac{h}{p} = \frac{hc}{[(2)(938.3 \text{ MeV})(10 \text{ MeV})]^{1/2}} = \frac{1240 \text{ MeV fm}}{[(2)(938.3)(10)(\text{MeV})^2]^{1/2}} = 9.05 \text{ fm} = 9.05 \times 10^{-15} \text{ m}$$

5-8 $\lambda = \frac{h}{p} = \frac{h}{(2mK)^{1/2}} = \frac{h}{(2meV)^{1/2}} = \left[\frac{h}{(2me)^{1/2}} \right] V^{-1/2}$
 $\lambda = \left[\frac{6.626 \times 10^{-34} \text{ Js}}{(2 \times 9.105 \times 10^{-31} \text{ kg} \times 1.602 \times 10^{-19} \text{ C})^{1/2}} \right] V^{-1/2}$
 $\lambda = \left[\frac{1.226 \times 10^{-9} \text{ kg}^{1/2} \text{ m}^2}{sC^{1/2}} \right] V^{-1/2}$

5-10 As $\lambda = 2a_0 = 2(0.0529) \text{ nm} = 0.1058 \text{ nm}$ the energy of the electron is nonrelativistic, so we can use

$$p = \frac{h}{\lambda} \text{ with } K = \frac{p^2}{2m};$$

$$K = \frac{h^2}{2m\lambda^2} = \frac{(6.626 \times 10^{-34} \text{ J} \cdot \text{s})^2}{2(9.11 \times 10^{-31} \text{ kg})(1.058 \times 10^{-10} \text{ m})^2} = 21.5 \times 10^{-18} \text{ J} = 134 \text{ eV}$$

This is about ten times as large as the ground-state energy of hydrogen, which is 13.6 eV.

5-11 (a) In this problem, the electron must be treated relativistically because we must use relativity when $pc \approx mc^2$. (See problem 5-5). the momentum of the electron is

$$p = \frac{h}{\lambda} = \frac{6.626 \times 10^{-34} \text{ J} \cdot \text{s}}{10^{-14} \text{ m}} = 6.626 \times 10^{-20} \text{ kg} \cdot \text{m/s}$$

and $pc = 124 \text{ MeV} \gg mc^2 = 0.511 \text{ MeV}$. The energy of the electron is

$$E = (p^2 c^2 + m^2 c^4)^{1/2}$$

$$= \left[(6.626 \times 10^{-20} \text{ kg} \cdot \text{m/s})^2 (3 \times 10^8 \text{ m/s})^2 + (0.511 \times 10^6 \text{ eV})^2 (1.602 \times 10^{-19} \text{ J/eV})^2 \right]^{1/2}$$

$$= 1.99 \times 10^{-11} \text{ J} = 1.24 \times 10^8 \text{ eV}$$

so that $K = E - mc^2 \approx 124 \text{ MeV}$.

(b) The kinetic energy is too large to expect that the electron could be confined to a region the size of the nucleus.